

## Non-legislated emissions and PN of two passenger cars with gasoline-butanol blends

*Increasing the sustainability of individual transportation and replacing a part of fossil energy in traffic by renewable energy carriers are worldwide important objectives. Bioalcohols are generally recognized as one of very useful alternatives. The global share of bioethanol used for transportation is continuously increasing. Butanol, a four-carbon alcohol, is considered in the last years as an interesting alternative fuel, both for diesel and for gasoline application. Its advantages for engine operation are: good miscibility with gasoline and diesel fuels, higher calorific value than ethanol, lower hygroscopicity, lower corrosivity and possibility of replacing aviation fuels.*

*In the present work, the emissions of two gasoline vehicles – with older and with newer technology – were investigated in dynamic-, stationary and cold start operation.*

Key words: *alternative fuels, alcohols, nanoparticles, ammonia, cold start, emissions gasoline*

### 1. Introduction

#### 1.1. Butanol and its effects on SI-engines

Butanol ( $\text{CH}_3(\text{CH}_2)_3\text{OH}$ ) has a four-carbon structure and is a higherchain alcohol than Ethanol, as the carbon atoms can either form a straight chain (n-butanol) or a branched structure (iso-butanol), thus resulting in different properties. Consequently, it exists as different isomers depending on the location of the hydroxyl group (-OH) and carbon chain structure, with butanol production from biomass tending to yield mainly straight chain molecules. 1-butanol, better known as *n*-butanol (normal butanol), has a straight-chain structure with the hydroxyl group (-OH) at the terminal carbon.

*n*-butanol is of particular interest as a renewable biofuel as it is less hydrophilic, and possesses higher energy content, higher cetane number, higher viscosity, lower vapour pressure, higher flash point and higher miscibility than ethanol, making it more preferable than Ethanol for blending with diesel fuel. It is also easily miscible with gasoline and it has no corrosive, or destructing activity on plastics, or metals, like ethanol or methanol.

Several research works were performed with different butanol blends BuXX [1–9]. Generally, there are advantages of higher heat value (than ethanol). The oxygen content of butanol has similar advantages, like with other alcohols: tendency of less CO & HC, but possibility of increasing  $\text{NO}_x$  (depending on engine parameters setting).

The good miscibility, lower hygroscopicity and lower corrosivity make butanol interesting alternative.

The trend of downsizing the SI-engines in the last years implies much higher specific torques and with it an aptitude of knocking and mega-knocking at high- and full load. The alcohols have a higher octane numbers (RON), are more resistant to knocking and are a welcomed solution for this new technology of engines [1].

A basic research of butanol blends Bu20 & Bu100 was performed on mono-cylinder engines with optical access to the combustion chamber [2, 3]. One of the engines was with GDI configuration. It was demonstrated, that the alco-

hol blend improved the internal mixture preparation and reduced the carbonaceous compounds formation and soot.

Concerning the characteristics of combustion Bu100 was similar to gasoline. This research considered only little number of constant operating points.

Using *n*-butanol in an optical port fuel injection (PFI) SI engine slightly higher combustion rates and lower formation of particulates was found compared to gasoline [4, 5]. Similarly [6] reported that the duration of the early combustion stage and length of combustion in an SI engine were, compared to gasoline, shortened with increased *n*-butanol share, and slightly lower variability of indicated mean pressure (IMEP) was observed when running on neat *n*-butanol. Shorter early combustion stage, faster combustion and better combustion stability were also observed by other researchers [7, 8].

The alcohol blend fuels E85 & Bu85 were tested on a vehicle with TWC in road application and with on-board measuring system for exhaust emissions [9]. It was stated for butanol, that it has no significant influence on CO & HC, but it increases strongly  $\text{NO}_x$ . Nevertheless, this is due to the limits of lambda regulation and as effect of it to the production of too many lean lambda excursions during the transients.

The warm operation with Bu85 was with no problems, the cold startability and emissions were not investigated.

#### 1.2. Non-legislated emissions of gasoline cars

The most important non-legislated emission components in present discussions are: the nanoparticles (NP), ammonia ( $\text{NH}_3$ ), nitrous oxide ( $\text{N}_2\text{O}$ ), formaldehyde (HCHO) and acetaldehyde (MeCHO).

The nanoparticles (NP) became an important research topic, since the first introduction of legal nanoparticle counts limits (Euro 5b) for DI SI passenger cars in EU beginning of 2013. In this situation, the NP and especially the metal oxides emissions from additive packages of lube oils and fuels, become an important topic for all kinds of engines. Lube oil contributes to the NP-emission especially

at cold start [10–12]. These new aspects have to be investigated with Ethanol blend fuels Exx.

Further gaseous substances, which may be present under certain conditions in very low concentrations in the exhaust gases are considered to be potential candidates for future legal limitation. These non-legislated emission components are: ammonia (NH<sub>3</sub>), nitrogen dioxide (NO<sub>2</sub>), nitrous oxide (N<sub>2</sub>O), formaldehyde (HCHO) and acetaldehyde (MeCHO) – all of them quite easy to be measured and indicated with FTIR.

Production of ammonia (NH<sub>3</sub>) in the exhaust of gasoline cars with TWC was demonstrated in [13] and [14] – this especially at transient operation with rich excursions of lambda. The other components were little investigated in connection with E85-operation. From the research of the authors can be stated, that with a correctly working TWC (at warm operation) there are usually no measurable concentrations of NO<sub>2</sub> and N<sub>2</sub>O and the HCHO – values show a noise below 1 ppm [15]. The components HCHO and MeCHO are supposed to produce a peak at cold start, but are little investigated and presented in the literature.

The presented tests were performed in the IC-Engines Laboratory of the University of Applied Sciences, Biel, CH within the framework of project GasBut (gasoline + butanol).

This research was conducted on two cars: an older one, with MPI & λ = 1 concept and a newer one (Euro 5), with GDI, λ = 1 concept and flex fuel aptitude.

The test vehicles were driven at WLTC cold & warm, as well as at a steady state cycle (SSC). The measurements of legislated and non-legislated emissions (NP & FTIR) were attached.

Special attempts of cold starts were conducted and compared with the equivalent results with Bu0 & Exx. The tests were performed with Bu0, B15 and Bu30.

This research enabled a complete insight in the non-legislated emissions at cold start and in repetitive transient operation with quite different state of the gasoline cars.

## 2. Test vehicles, fuels and lubricants

The tests on gasoline vehicles were performed: with a Renault 18 Break (SI, MPI, TWC), which represents an older technology in this project and with a flex fuel vehicle (FFV) Volvo V60 (GDI, Euro 5), which represents a newer technology. These vehicles were operated with gasoline, in original condition (TWC) and with two butanol blend fuels Bu15 and Bu30. The vehicles are presented in Fig. 1 and Tab. 1.



Fig. 1. Gasoline vehicles for research of emissions

Table 1. Data of tested vehicles

Vehicle	Renault 18 Break	Volvo V60 T4F
Engine code	J7T-718	B4164T2
Number and arrangement of cylinders	4 / in line	4 / in line
Displacement cm <sup>3</sup>	2164	1596
Power kW	74 @ 5000 rpm	132 @ 5700 rpm
Torque Nm	162 @ 2000 rpm	240 @ 1600 rpm
Injection type	MPI	DI
Curb weight kg	1110	1554
Gross vehicle weight kg	1585	2110
Drive wheel	Front-wheel drive	Front-wheel drive
Gearbox	m5	a6
First registration	01.04.1985	27.01.2012
Exhaust	EURO 0	EURO 5a
VIN	VF1135B00F0000505	YV1FW075BC1043598
Vehicle	Renault 18 Break	Volvo V60

### 2.1. Fuels

The gasoline used was from the Swiss market, RON 95, according to SN EN228; n-butanol was purchased from Thommen-Furler AG. As blend fuels were used: Bu15 and Bu30 (15% vol. and 30% vol. butanol).

Table 2. Fuel properties of the test fuels

Specification		RON 95	n-butanol
Other name		Gasoline, Bu0	1-butanol
Formula		-	C <sub>4</sub> H <sub>10</sub> O
Density	[kg/dm <sup>3</sup> ]	0.737	0.810
Stoichiometric AF-ratio	[kg air]	14.70	11.10
Lower heating value	[MJ/kg]	42.7	33.0
O <sub>2</sub> fraction	[% <sub>m</sub> ]	1.70	21.62
Boiling range	[°C]	38-175	115-119
Blending RON		95	99
Blending MON		87	84
Self-ignition temperature	[°C]	300	343
Flash point	[°C]	<-40	34
Viscosity @ 40°C	[mPa*s]	0.83	2.9
		<b>Bu15</b>	<b>Bu30</b>
Density	[kg/dm <sup>3</sup> ]	0.748	0.759
Stoichiometric AF-ratio	[kg air]	14.12	13.55
Lower heating value	[MJ/kg]	41.1	39.6
O <sub>2</sub> fraction	[% <sub>m</sub> ]	3.50	8.08

Table 2 presents the most important data of the fuels (according to the literature sources). It can be remarked that with increasing share of butanol the oxygen content of blend fuel increases and the heat value and stoichiometric air requirement decrease.

## 2.2. Lubricants

In the present tests the lube oil was not changed and analyzed – the same oil was used for all tests.

## 3. Test methods and instrumentation

### 3.1. Chassis dynamometer and standard test equipment

- roller dynamometer: Schenk 500 GS 60
- driver conductor system: Tornado, version 3.3.
- CVS dilution system: Horiba CVS-9500T with Roots blower
- air conditioning in the hall automatic (intake- and dilution air)

The driving resistances of the test bench were set according to the legal prescription.

### 3.2. Test equipment for regulated exhaust gas emissions

This equipment fulfils the requirements of the Swiss and European exhaust gas legislation.

- gaseous components:  
exhaust gas measuring system Horiba MEXA-9400H  
CO, CO<sub>2</sub> – infrared analysers (IR), HCIR, HCFID, NO/NO<sub>x</sub>, CLA

The dilution ratio DF in the CVS-dilution tunnel is variable and can be controlled by means of the CO<sub>2</sub>-analysis.

### 3.3. FTIR

FTIR (Fourier Transform Infrared) Spectrometer (AVL Sesam) offers the possibility of simultaneous, time-resolved measurement of approx. 30 emission components – among others: NO, NO<sub>2</sub>, NO<sub>x</sub>, NH<sub>3</sub>, N<sub>2</sub>O, HCN, HNCO, HCHO and MeCHO.

### 3.4. Nanoparticle analysis

The measurements of NP size distributions were conducted with different SMPS-systems, which enabled different ranges of size analysis:

SMPS: DMA TSI 3081 and CPC TSI 3772 (9.8–429 nm)

nSMPS: nDMA TSI 3085 and CPC TSI 3025 (3–64 nm).

For the dilution and sample preparation an ASET system from Matter Aerosol was used, Fig. 2 (ASET – aerosol sampling and evaporation tube). This system contains:

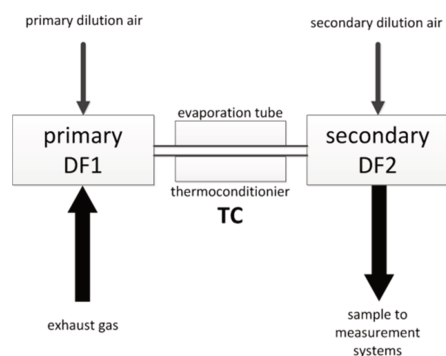


Fig. 2. Set-up of dilution stages and sample preparation for nanoparticle measurements

- Primary dilution air – MD19 tunable minidiluter (Matter Eng. MD19-2E)
- Secondary dilution air – dilution of the primary diluted and thermally conditioned measuring gas on the outlet of evaporative tube.

- Thermoconditioner (TC) – sample heating at 300°C.

The measuring set-up on chassis dynamometer and the sampling positions for particle analytics are represented in Fig. 3.

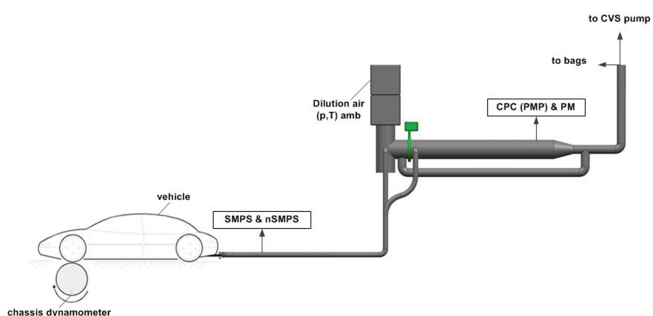


Fig. 3. Sampling of exhaust gas for analysis of particles

## 4. Test procedures on chassis dynamometer

The vehicles were tested on a chassis dynamometer in the dynamic driving cycles WLTC and at constant speeds in the steady state cycle SSC. SSC consists of 20 min steps at constant vehicle speeds 95, 45 km/h and idling, which are driven from the highest to the lowest speed. These vehicle speeds respond to the average speeds in parts of the WLTC. The test sequences with all fuels were identical: WLTC with cold start (20–25°C), 10 min idling for bag evaluation, acceleration to 95 km/h and continuation of the SSC.

In terms of the driving cycles an approach to find a homogenized world-wide driving cycle was finished with the development of the homogenized WLTP world-wide light duty test procedure. The WLTC (world-wide light duty test cycle) represents typical driving conditions around the world and is developed based on combination of collected in-use data and suitable weighting factors. This cycle has been used also in this study, Fig. 4. It represents different driving situation, like city, over-land and speed-way.

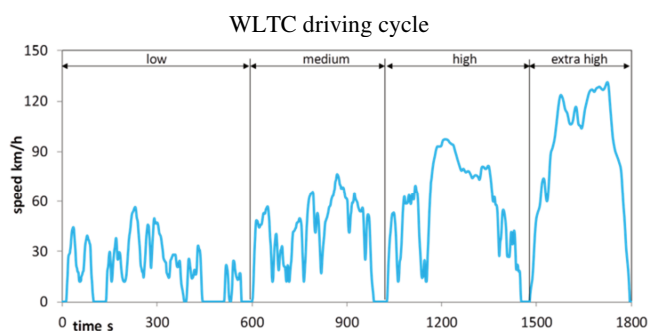


Fig. 4. WLTC driving cycle

## 5. Results

### 5.1. Comparison of emissions of vehicles with older and with newer technology

Regarding the comparison of emissions time-plots in WLTC (not represented here), it can be generally remarked for all three fuels (Bu0, Bu15 and Bu30):

- with the older vehicle (R18) there are considerably higher emissions of CO and HC at cold start and there are higher and more frequent peaks of all components (CO, HC and NO<sub>x</sub>) during the driving cycle,
- all non-legislated emissions: NH<sub>3</sub>, HCHO, MeCHO and N<sub>2</sub>O are for R18 significantly higher.

Considering the integral average emissions in WLTC (whole cycle), Figs 5 and 6, these statements can be confirmed:

- higher CO- and HC-values with R18,
- with Bu15 CO is reduced more for V60, than for R18,
- with Bu30 CO for V60 stays at the level of Bu15, while for R18 it increased again to the original level of Bu0,
- HC for both vehicles is unchanged, or slightly reduced with Bu15, but it generally increases with Bu30,
- NO<sub>x</sub> is strongly increased by both BuXX fuels for the older vehicle (R18) and it is reduced for the newer ve-

hicle (V60) – this is a sensitive indication of better functioning of the Lambda regulation of V60, with less “lean-excursions”,

- the nanoparticle emission of V60 is significantly reduced with both BuXX-fuels; the PN emission of R18 is not influenced by the fuel,
- all non-legislated emissions: NH<sub>3</sub>, HCHO, MeCHO and N<sub>2</sub>O are for R18 significantly higher,
- there is a tendency of increasing HCHO and MeCHO with increasing BuXX for both vehicles,
- with increasing BuXX there is an increase of NH<sub>3</sub> for V60 and approximately no influence for R18.

One example of time-plots of non-legislated gaseous components, with both vehicles and with gasoline (Bu0), is given in Fig. 7. It clearly demonstrates the advantages of the newer car (V60).

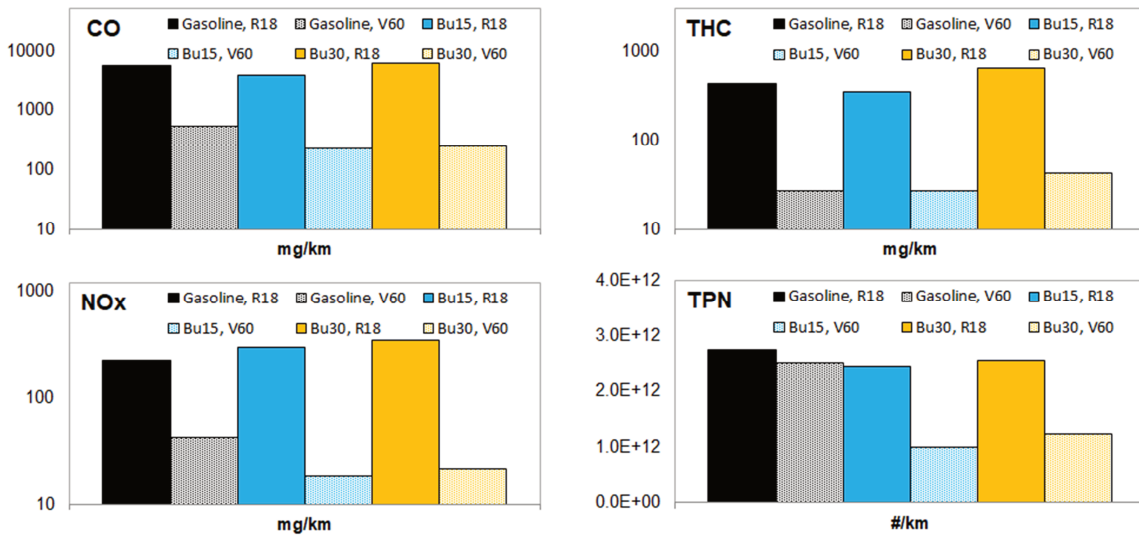


Fig. 5. Comparisons of emissions R18 vs V60 in WLTC cold with Bu0, Bu15 & Bu30

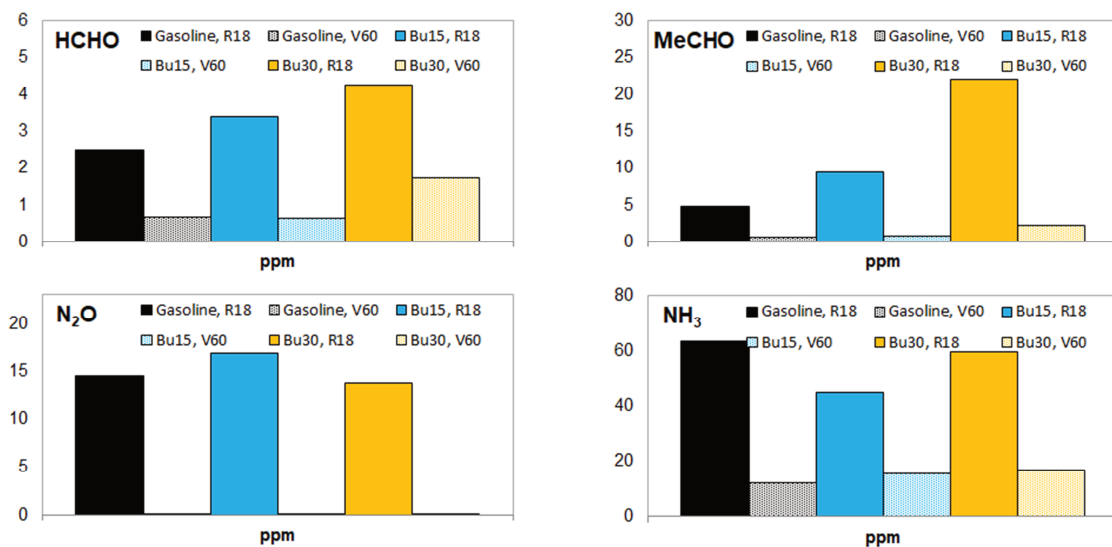


Fig. 6. Comparisons of non-legislated emissions in WLTC cold

Fig. 8 illustrates the relationships of emissions at 95 km/h (in the 1st step of SSC).

A look on the average emission values in SSC allows the general statements:

- in most cases there are higher CO-and HC-values for R18,
- with increasing Bu-content at 95 km/h there is a strong increase of NO<sub>x</sub> for R18 and no influence on NO<sub>x</sub> for V60,
- the nanoparticle emission of V60 is significantly reduced with both BuXX-fuels; the PN emission of R18 is not influenced by the fuel,
- in most cases the higher values of NH<sub>3</sub>, N<sub>2</sub>O and MeCHO are confirmed for R18.

up to 60 ppm and MeCHO up to 220 ppm. N<sub>2</sub>O emission peaks depend only few from the fuel variant. NH<sub>3</sub>-values are generally low after the cold start and they become higher in the hot last part of the cycle.

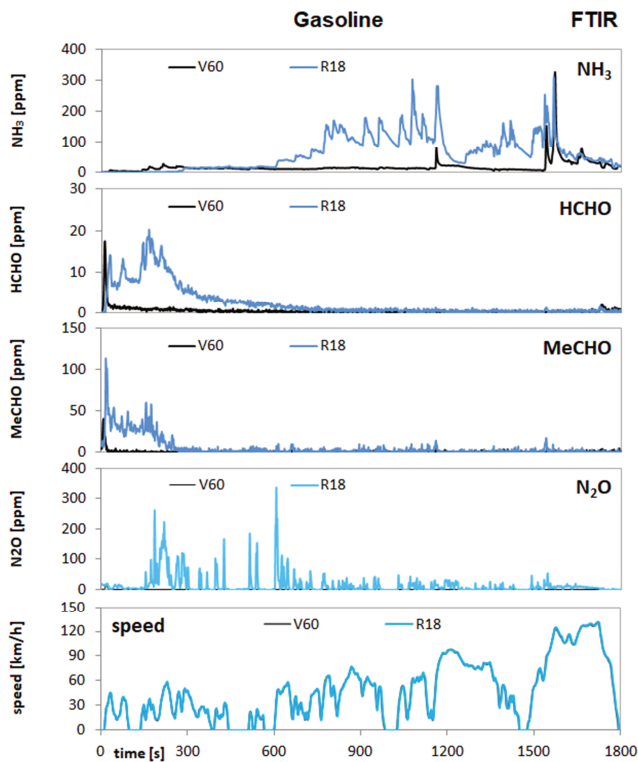


Fig. 7. Comparison of NH<sub>3</sub>-, HCHO-, MeCHO and N<sub>2</sub>O-emissions of two vehicles during the driving cycle WLTC cold

In the first step (95 km/h) Volvo (V60) has with gasoline higher nanoparticle emissions (CPC), than Renault (R18). With Bu15 and Bu30 this is no more the case, since the NP are for V60 considerable reduced with BuXX.

After switching the operation to idling there is for R18 an increase of NP (CPC), because there are the highest PN-emissions at idling for this vehicle. These NP consist in a large portion of unburned lube oil and it is not surprising that their number increases gradually with the cooling down the exhaust system and the catalyst (not represented here).

The highest NP-emissions at idling of R18, as well as their appearance mainly in the nuclei mode are documented in Fig. 9. The nanoSMPS offers at certain operating points, especially at 45 km/h, valuable supplementary information.

### 5.2. Non-legislated emissions of both vehicles

Figures 10 & 11 represent for both cars some non-legislated components in the first part of the cycle with cold start and warm-up. The sequence of increased emission peaks with higher Bu-content is clearly repetitive. There are considerable peak values with Bu30. For R18: HCHO up to 30 ppm and MeCHO up to 950 ppm and for V60: HCHO

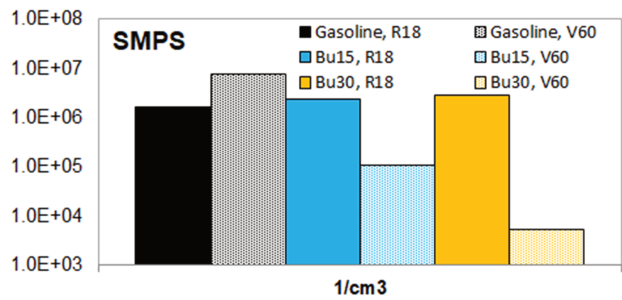
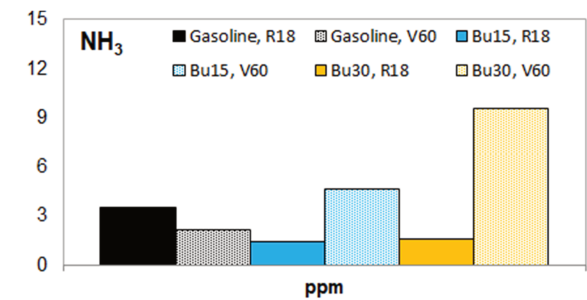
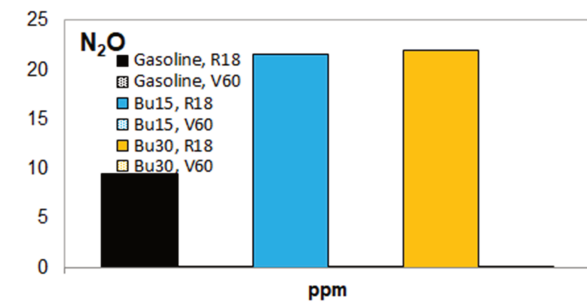
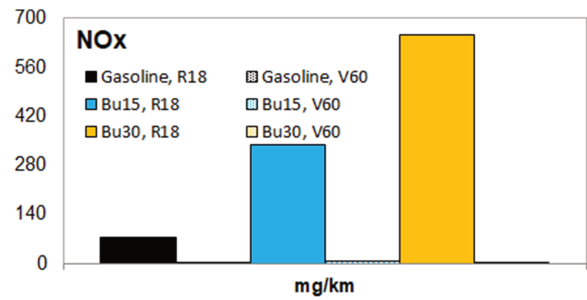


Fig. 8. Comparison of exhaust emissions of two vehicles at 95 km/h with different fuels

Figures 12 & 13 offer a consideration of SMPS particle size distributions for both vehicles, with three fuels and in all steps of the SSC.

For R18, the particle size distributions with SMPS (and with nSMPS) show principally higher PN-values with higher butanol content. At 45 km/h there is a major part of nanoparticles in the smallest sizes, below the measuring range of SMPS. The highest PN-concentration are reached at idling. This vehicle is known to produce excessive NP-emissions in nuclei mode, which originate from the higher lube oil consumption.

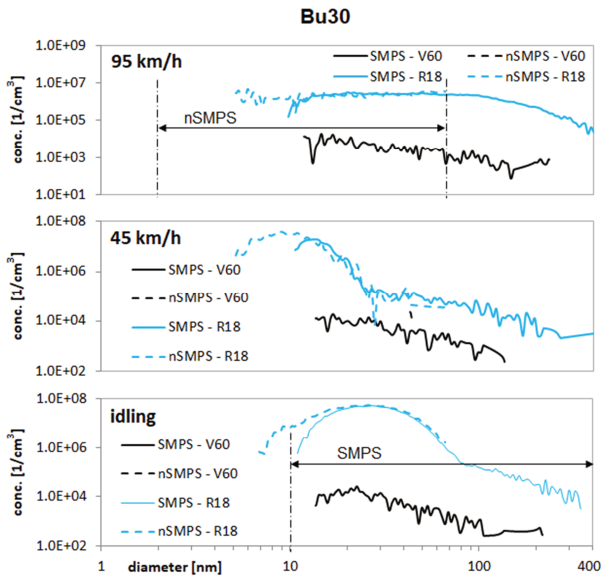


Fig. 9. Particle Size Distribution (PSD) during the SSC cycle. Comparison SMPS – nSMPS of two vehicles

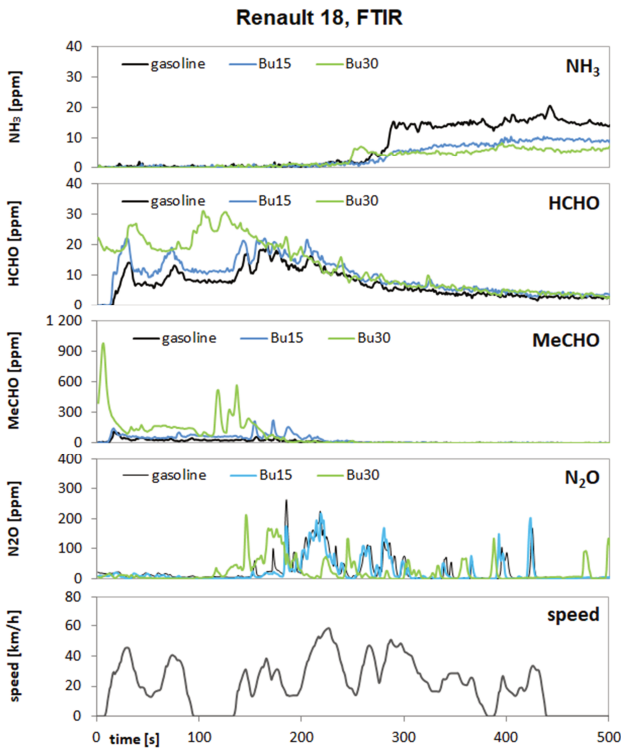


Fig. 10. Comparison of NH<sub>3</sub>-, HCHO-, MeCHO- & N<sub>2</sub>O-emissions in the first part of WLTC cold with different fuels

For V60, there is an inverse influence of Bu-blends: there is a clear lowering of particle number (PN) with increasing BuXX. At idling, generally the lowest PN counts concentrations are resulting.

From the comparisons in this section, it can be concluded that the different engines' ages and technology (different mixtures' preparations MPI/DI, combustion, lube oil consumption and exhaust aftertreatment) have a significant impact on the emissions and especially on the emissions at cold start.

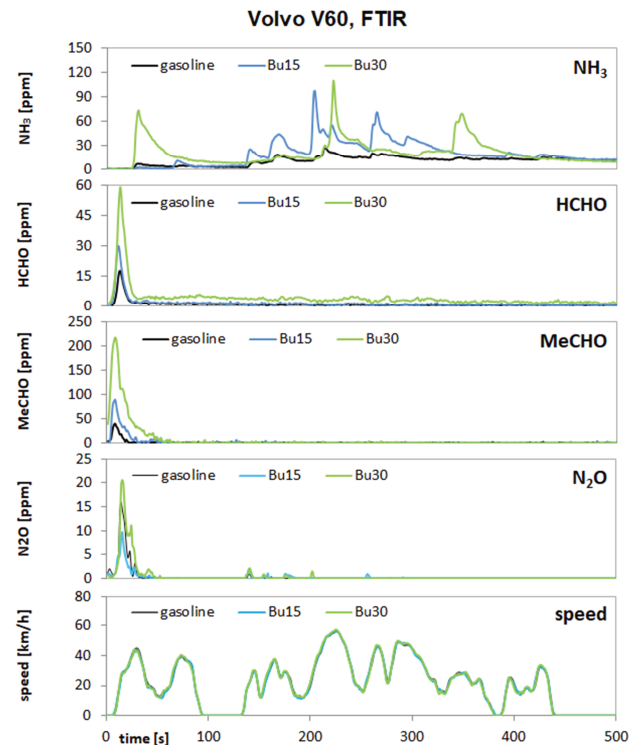


Fig. 11. Comparison of NH<sub>3</sub>-, HCHO-, MeCHO- & N<sub>2</sub>O-emissions in the first part of WLTC cold with different fuels

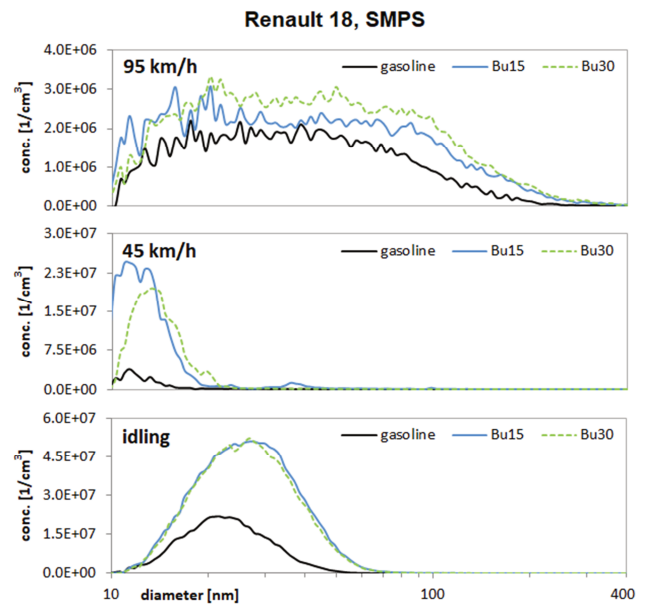


Fig. 12. Comparison of the Particle Size Distribution (PSD) during the driving cycle SSC with different fuels

### 5.3. Cold start

Repetitive cold start tests were performed with Volvo V60 and with Bu0/Bu15/Bu30. For cold starts (CS), two ranges of start temperature were considered: summer cold start (20 to 25°C, conditioning in the test hall), or mild winter cold start (-2 to 4°C, conditioning outside in the cold weather period). For simplification of titles and descriptions these temperature ranges will be designed, as 20°C and 0°C.

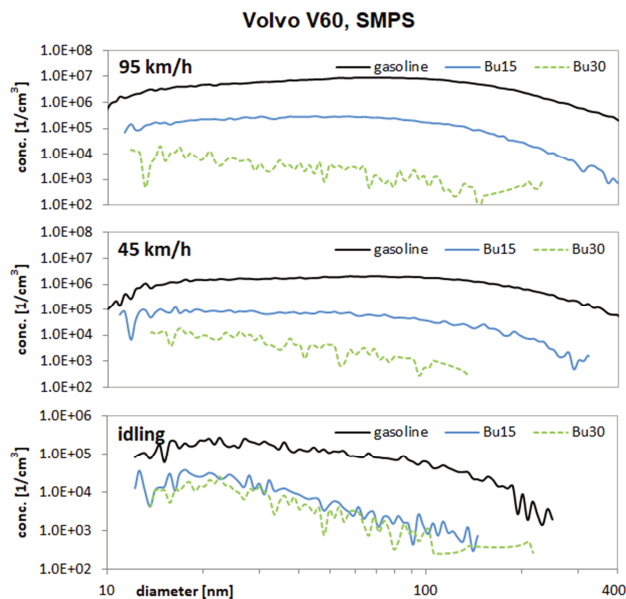


Fig. 13. Comparison of the Particle Size Distribution (PSD) during the driving cycle SSC with different fuels

In the preliminary tests with gasoline two variants of cold start were investigated:

- a. cold start at idling (without chassis dynamometer),
- b. cold start with acceleration to 20 km/h and  $v = \text{const} = 20 \text{ km/h}$  on the chassis dynamometer, the braking resistances were set according to legal prescriptions and they responded to the horizontal road.

It was stated after this test period, that the CS on chassis dynamometer (with 20 km/h) does not bring any further information potentials and further research was generally limited to the CS at idling. Vehicle, which was conditioned outside for the mild winter CS was pushed in the test hall, attached to the measuring systems, started and operated in the conditions of the hall (intake air 20–25°C). After the test, the vehicle was conditioned by driving a NEDC on the chassis dynamometer.

Fig. 14 shows some non-legislated gaseous components, comparing Bu0/Bu15/Bu30 in two temperature domains of the CS: 0°C and 20°C. With higher Bu-content the peaks of formaldehyde HCHO and of acetaldehyde MeCHO increase. Starting with a lower temperature, these peak-values are higher and can attain for MeCHO 250 ppm. The ammonia NH<sub>3</sub> concentrations are at cold start (CS) near to zero and they increase slightly after engine warms up. Nevertheless, there is for NH<sub>3</sub> no correlation with fuel quality.

Fig. 15 compares the nanoparticle emissions with the fuels Bu0/Bu15/Bu30 at CS in both temperature ranges 0°C & 20°C. CPC (condensation particle counter) measures the particle numbers of all particle sizes according to the PMP-guidelines. SMPS (scanning mobility particle sizer) measures the particle numbers in function of their size.

The SMPS-particle size distributions were taken in the successive parts of the warm-up period: (1) 0–120 s; (2) 120–300 s and (3) 300–600 s.

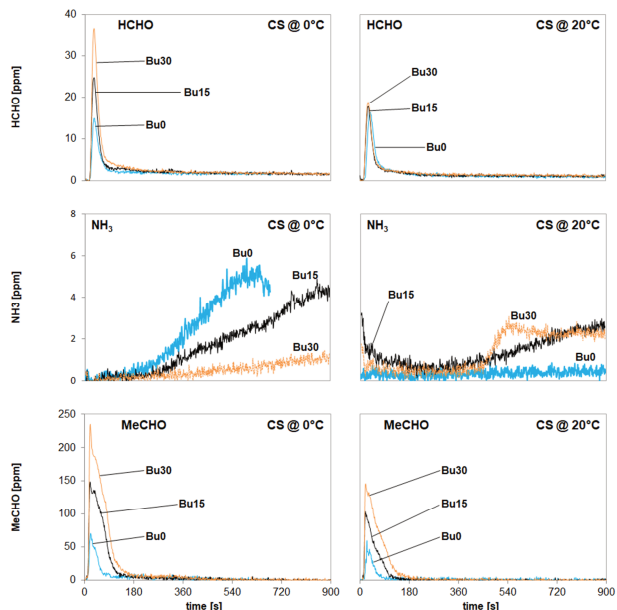


Fig. 14. Comparison of the non-legislated gaseous emissions during cold start at idling with different fuels, measured with FTIR at tailpipe

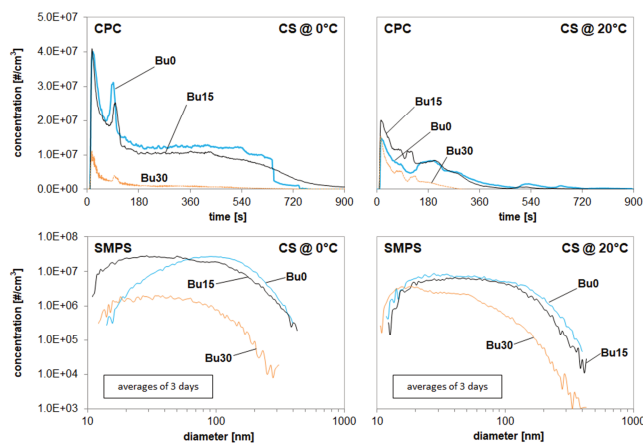


Fig. 15. Comparison of the particles counts during cold start at idling with different fuels, measured with both systems at tailpipe

The successive SMPS-scans of each CS-attempt (not represented here) showed clearly the lowest PC-level of the latest sample. The 1<sup>st</sup> sample was well repeatable and the PSD's in Fig. 4 are averages from three cold starts of the 1<sup>st</sup> scan (in the period 0–120 s).

The CPC-signals at 0°C have a second peak after approximately 2 min. This is visible particularly with gasoline (E0). This peak is a repeatable event, it can also be found in other emission courses (like N<sub>2</sub>O) and it is attributed to the changes introduced by the engine ECU in function of temperature, like possibly catalyst heating, switching of internal EGR by vario cams, or heat management.

The most important information of Fig. 15 is that Bu15 emits similar level of particle counts concentration, like Bu0, while B30 reduces clearly the PN emissions. Bu15 has similar oxygen content like E10. Nevertheless, it was found that Bu15 produces significantly higher peaks of MeCHO and HCHO at cold start than E10 [3].

## 6. Conclusions

The elaborated results allow following observations for R18:

- At cold start and warm-up all three investigated fuels produce increased CO-, HC- and NP-values in similar way.
- The emissions of HCHO and MeCHO at cold start increase in the sequence of increasing butanol content.
- In the “high” and “extra high” parts of WLTC there are the highest NH<sub>3</sub> peaks, which coincide with the strongest acceleration events in the cycle.
- Regarding the average emission values in WLTC cold: with increasing butanol content (BuXX) there is a clear tendency of increasing the emissions of: NO<sub>x</sub>, HCHO, MeCHO and ETOH. The average emissions of N<sub>2</sub>O and NH<sub>3</sub> are independent on the BuXX.
- At steady state operation (in SSC) with increasing butanol content there are:
  - higher NO<sub>x</sub>-values at the highest speed (95 km/h),
  - higher PN-values at all operating conditions.
- With higher butanol content, the lambda regulation of this vehicle has difficulty to compensate the higher oxygen content of the fuel. As a result, there is a leaner operation and lower NO<sub>x</sub>-conversion in the TWC.
- Higher butanol content interferes more with the lube oil and tendentially increases the nanoparticles counts.
- Higher butanol content also creates favourable conditions to produce more formaldehyde (HCHO) and acetaldehyde (MeCHO) at cold start.

With B30 the excessive leaning was remarkable as a less powerful load responses and worse driveability. B30 is regarded as a maximum of butanol content to be recommended for this vehicle.

For Volvo V60 and for transient operation in WLTC can be remarked:

- With increasing portion of butanol in fuel (BuXX) there are increasing peak values of HC, HCHO, MeCHO, ETOH and N<sub>2</sub>O at cold start.
- During and after the acceleration events in the highest part of the cycle there are emission peaks of some components, but they cannot be attributed to a specific butanol content (BuXX).
- The comparison of average emission values in WLTC, confirms the lower CO- and lower PN-values with BuXX, while it is difficult to notice the difference between Bu15 and Bu30.

- The average of FTIR-values confirms the higher values of: HCHO, MeCHO and NH<sub>3</sub> with BuXX.
- There is a clear lowering of particle number (PN) with increasing BuXX.

### Comparison R18-V60 in WLTC

- Higher CO- and HC-values with R18 and no clear influence of fuel on these emissions.
  - HC for both vehicles is unchanged, or slightly reduced with Bu15, but it generally increases with Bu30.
  - NO<sub>x</sub> is strongly increased by both BuXX fuels for the older vehicle (R18) and it is reduced for the newer vehicle (V60) – this is a sensitive indication of better functioning of the lambda regulation of V60, with less “lean-excursions”.
  - The nanoparticle emission of V60 is significantly reduced with both BuXX-fuels; the PN emission of R18 is not influenced by the fuel.
  - All non-legislated emissions: NH<sub>3</sub>, HCHO, MeCHO and N<sub>2</sub>O are for R18 significantly higher.
  - There is a tendency of increasing HCHO and MeCHO with increasing BuXX for both vehicles.
  - With increasing BuXX there is an increase of NH<sub>3</sub> for V60 and approximately no influence for R18.
- For cold start tests with Volvo V60 can be concluded:
- With increasing butanol content (Bu0/Bu15/Bu30) the emissions at cold start are influenced in following way:
    - Higher peaks of acetaldehyde (MeCHO) at start,
    - Higher peaks of formaldehyde (HCHO) at start,
    - The nanoparticles with Bu15 have similar level as with Bu0 (both CPC and SMPS), with Bu30 they are approximately 1 order of magnitude lower.
  - The higher temperature of the cold start generally lowers the emission peaks.

It is important to mention that the original plans of this project part were to test the cold start with Bu85. This was also tried in both temperature domains (0°C & 20°C) but without success. The start and the operation were not possible with this FFV. Butanol has a higher boiling point, than ethanol and therefore the quality of mixture preparation (part of evaporated fuel) with butanol is worse. The investigated vehicle (FFV) is developed for ethanol and cannot work adequately with higher butanol contents.

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## Nomenclature

AFHB	Abgasprüfstelle FH Biel, CH
ASET	Aerosol Sampling and Evaporation Tube
ASTRA	Amt für Strassen (CH)
BAFU	Bundesamt für Umwelt, (Swiss EPA)
BfE	Bundesamt für Energie
Bu	butanol
BuXX	butanol content XX
CLA	chemiluminescent analyzer
CLD	chemiluminescent detector

CO	carbon monoxide
CO <sub>2</sub>	carbon dioxide
CPC	condensation particle counter
CS	cold start
CVS	constant volume sampling
DF	dilution factor
DI	direct injection
DMA	differential mobility analyzer
ECU	electronic control unit



EV	Erdölvereinigung	N <sub>2</sub> O	nitrous oxide
FFV	flex fuel vehicle	N <sub>2</sub>	nitrogene
FID	flame ionisation detector	NO <sub>x</sub>	nitric oxides
FTIR	Fourier Transform Infrared analyzer	NP	nanoparticles < 999 nm
GasBut	gasoline buthanol project	PC	particle counts (integrated)
HC	unburned hydrocarbons	PN	particle numbers
HCHO	formaldehyde	PSD	particle size distribution
HCN	hydrogen cyanide	R18	Renault 18
HNCO	isocyanic acid	SMPS	scanning mobility particle sizer
MD	minidiluter	TC	thermoconditioner
MeCHO	acetaldehyde	TWC	three way catalyst
NEDC	New European Driving Cycle (ECE + EUDC)	V60	Volvo V60
NH <sub>3</sub>	ammonia	WLTC	worldwide harmonized light duty test cycle
NO	nitrogen monoxide	WLTP	worldwide harmonized light duty test procedure
NO <sub>2</sub>	nitrogen dioxide	3WC	three way catalyst

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