

Research of the Impact of the Heating Rate on Adhesion of Scale Arising in the Process of Heating of the Steel Charge Before the Plastic Reworking

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INTRODUCTION

A plastic processing of hot is preceded by heating the steel charge. The heated charge, contacting with the atmosphere of the working chamber, undergoes oxidation process, resulting in a scale. The presence of scale layer on the surface of the charge is an important issue for both the heating process and subsequent plastic processing. Currently in every area of the economy, including the steel industry, there is a desire to improve the efficiency of production processes, including by reducing the consumption of heat and the substantial reduction. One of the elements contributing to the maintenance of low rates of specific heat consumption and minimal loss of steel scale in the process of heating a charge prior to plastic processing, is the selection of appropriate heating technology. This problem has been widely described in the paper ((Neumann, 1971, Kieloch, 2002, Boryca et al., 2006). Ensuring the quality of semi-finished steel and reduce device failure, both heating and rolling, is conditional on achieving sufficient adhesion of scale during the heating process, and after leaving the furnace by steel charge. The behavior of adhesion of scale to the proper level is important both for the production of sheet metal (Kieloch, 2002, Boryca et al., 2006, Lazič et al., 2008), as well as in the manufacture of steel (Lee & Kim, 2013, Higginson et al., 2002, Stanković et al., 2012, Bajić et al., 2008, Varga et al., 2007). It can be concluded that the next step in optimizing the heating process should be to development technologies limiting the loss of steel, which simultaneously would guarantee the such adhesion to the steel substrate to mill scale does not fall off in the furnace and was easily removed after leaving it. Previous studies of adhesion of scale related to the effect of temperature and heating time, the furnace atmosphere and the condition of the surface charge (Boryca et al., 2006, Boryca et al., 2015, Tyusenkov & Cherepashkin, 2014).

THEORETICAL BASIS

Scale adhesion to steel substrate is an important element to the selection of heating technology. The value of adhesion is determined by the mechanism of the formation of scale, which is described in the papers (Boryca et al., 2006). One of the parameters of heating technology that determines the consumption of heat and loss of steel by a scale, is the intensity of the increase of the surface temperature of the heated charge (Boryca et al., 2006, Varga et al., 2008, Varga et al., 2009, Wild et al., 2009). Correlation between the amount of the resulting scale of its adhesion can lead to the conclusion that it is necessary to consider the effect of heating rate on the adhesion (Boryca, 2007). The scale occurring in the preheating stage will affect the speed of the process during soaking. The loss of steel in the whole process will therefore be the sum of the losses of the two periods (stages). The loss of steel in the whole heating process is calculated from the equation (Kieloch, 2002):

$$z^N = z_p^N + z_w^N \tag{1}$$

or:

$$z^{N} = z_{p}^{N} + A_{w} \alpha_{w}^{C} \tau_{w} \exp\left(-\frac{D}{T_{w}}\right)$$
⁽²⁾

The loss of steel during preheating in the case of a linear increase of the surface temperature can-calculate from the equation (Kieloch et al., 2006):

$$z_p^N = \left[\frac{A\alpha_p^C \int_{T_0}^{T_p} exp\left(-\frac{D}{T}\right) dT}{M} + A\alpha_w^C \tau_w exp\left(-\frac{D}{T_w}\right)\right]^B$$
(3)

As soaking proceeds with constant temperature of the surface, the steel loss during this period may be calculated from the equation (Kieloch et al., 2006):

$$z_{w_{W}}^{N} = A' \tau_{w}^{B} \alpha_{w}^{C} \exp\left(-\frac{D}{T_{w}}\right)$$
(4)

or (Kieloch et al., 2006):

$$z_w^N = 0.405 A \alpha_w^C \exp\left(-\frac{D}{T_w}\right) \frac{s^2}{a_w} ln\left(0.516 \frac{Ms^2}{a_p} \frac{1}{\Delta t_k}\right)$$
(5)

Thus, with increasing of the heating rate M, the steel loss will decrease during the period of preheating and grow during soaking. Finally the equation is obtained (Kieloch, 2002):

$$z^{N} = \left[\frac{A\alpha_{p}^{C}\int_{T_{0}}^{T_{p}}exp\left(-\frac{D}{T}\right)dT}{M} + 0.405A\alpha_{w}^{C}\frac{s^{2}}{a_{w}}exp\left(-\frac{D}{T_{w}}\right)ln\left(0.516\frac{Ms^{2}}{a_{p}}\frac{1}{\Delta t_{k}}\right)\right]^{B}$$
(6)

The developed method of calculation of steel loss for two-stage heating allows the calculation for any number of heating stages. Numerical calculation of the steel loss depending on the rate of heating made for the model of pusher furnace (Kieloch, 2002, Golis et al., 1991) have demonstrated that loss of the steel as a function of preheating rate can be described by the following equations (Kieloch et al., 2006):

$$z = \frac{341.8}{M} + 0.781 \cdot \ln(0.0406 \cdot M) \tag{7}$$

$$z = 3,051 + 7,375 \cdot exp(-0,0162 \cdot M) \tag{8}$$

$$z = 2,862 \cdot exp(0,00011 \cdot M) + 6,0622 \cdot exp(-0,0132 \cdot M)$$
(9)

$$z = 2,674 + \frac{171,876}{M} \tag{10}$$

The results of studies (Kieloch, 2010) indicate that there is a rational rate of preheating, allowing to achieve a minimum loss of steel scale and maximum reduction of heat consumption. It is anticipated that similar results can be achieved when considering the scale adhesion to the steel substrate.

MEASUREMENT SYSTEM

The heating process was undertaken in an electric tube furnace (with combustion chamber) equipped with a programmable controller SHIMADEN-FP93 (Fig. 1).

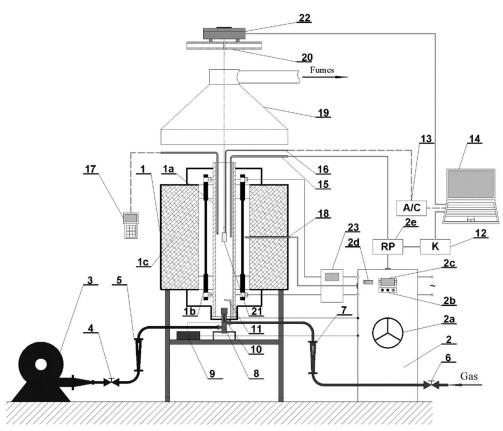


Fig. 1 Test stand scheme:

 Electric tube furnace, 1a) cylindrical chamber, 1b) resistance heating elements, 1c) thermal insulation, 2. The three-phase power supply cabinet, 2a) steering wheel to adjust the power settings of the furnace, 2b) desktop with additional features, 2c) keyboard and display of programmable controller, 2d) keyboard and display of the automatic security delimiter, 2e) programmable controller SHIMADEN – FP93, 3. Radial fan KONWEKTOR – WP-7,6, 4.
 Air valve, 5. Rotameter to measure the flow of the combustion air, 6. Gas valve, 7. Rotameter for the measure the flow of natural gas, 8. Gas burner, 9. Power transformer spark igniter, 10. Spark igniter, 11. Thermocouple flame control, 12. Converter YUKO – KO-485d, 13. A/D Card, 14. Notebook with software, 15. Control thermocouple PtRh10 (S), 16. Thermocouple PtRh10-Pt (S) or NiCr-Ni (K) to measure the temperature on the surface of the test sample, 17. Portable gas analyzer TESTO 350, 18. Control thermocouple, 19. Exhaust gas, 20. Isolated shelf for balance, 21. Test steel sample, 22. Electronic precision balance, 23. Electronic meter of electric energy

The process was carried out in two stages: preheating and soaking. The launch of the measuring system required previous programming of the heating characteristics using the controller. Determination of the heating characteristics relied on the amount of time steps and the maximum temperature for a given step. After running the program the furnace automatically realized the heating process in well-defined steps.

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The principle operation of the test stand has been presented in the article (Kieloch, 2002, Pilarczyk, 2006). The scale adhesion apparatus for descaling was designed and built. This is outlined in (Boryca et al., 2006).

METHODS OF ADHESION MEASURING

In order to achieve the research objectives, measurement methodology of adhesion for both hot and cold load was prepared. Mass method was applied, used for hot samples (Boryca et al., 2006). This method involves weighing samples in successive stages of the research. The adhesion of the scale layers can be expressed by mass scale rest after breaking into her total mass. The value of this adhesion determines the equation:

$$P_m = \frac{m_2 - m_3}{m_1 - m_3} \cdot 100\% \tag{11}$$

where:

m1 - mass of the sample after heating, kg,

m2 - mass of the sample after breaking the scale, kg,

m₃ - mass of the sample after cleaning, kg,

 P_{m} - adhesion of scale for the mass method, %.

Mass m_2 and m_3 are determined by weighing samples using an electronic balance WPS-360/C. To determine the mass m_1 is the equation:

$$m_1 = m_0 + \frac{z \cdot A}{0.74} \tag{12}$$

where:

m₀ - sample mass before heating, kg,

z - steel loss by scale, kg/m²,

A - sample surface, m².

In order to calculate the loss of steel equation was used:

$$z = \frac{m_0 - m_3}{A} \tag{13}$$

HEATING CURVES AND THEIR VERIFICATION

As presented above, carrying out experimental tests requires proper control of the furnace operation. This was done by controlling surface temperature of the sample. The temperature of the furnace chamber rises in the preheating process. The increase of the chamber temperature does not however evenly heat the charge throughout its volume. The charge is characterized by a specific value of thermal conductivity and a specified thickness. Therefore it require heating to a setpoint temperature for a limited period of time in the chamber at a constant temperature (soak period).

The increase of heating rate raises the temperature gradient over the crosssection of the charge. Therefore soaking time will increase with raising heating rate. Soaking times, for the designated heating tests, was found using computer programs to calculate the two-stage charge heating (preheating and soaking).For the designated heating curves the temperature of the sample surface and the heating chamber were measured. The results of calculations for an exemplary heating rate M = 200 K/h are shown in Fig. 2.

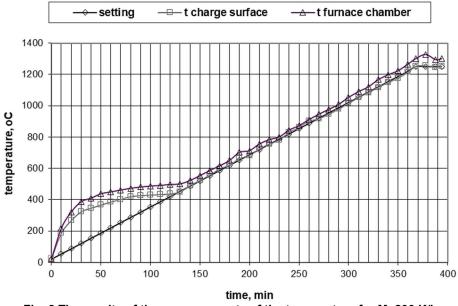


Fig. 2 The results of the measurements of the temperature for M=200 K/h

Analyzing Fig. 2, it can be concluded that the heating to a temperature of about 400°C is characterized by a non-linear increase in the temperature of the charge surface. Such situation is caused by the influence of thermal energy supplied from a gas burner whose power cannot be adjusted at steady flows of gas and combustion air. Once at the temperature of the sample surface greater than 400°C, this parameter is characterized by an almost linear nature of the increase. Deviation from the setting is approximately ±15°C, and the mean value of the deviation of + 2°C. Surface temperature deviation from the setting to a temperature of 400°C does not have a significant impact on the scale adhesion to the substrate steel, because in this area the scale formed with smaller quantities. The adhesion of scale at temperatures below 1000°C is at a level not exceeding 5% (Boryca et al., 2006). The temperature of the operating cell (heating) is maintained at an average level around 40°C above the temperature of the sample surface. The average increase in surface temperature of the sample is 33.2 K/10 min, giving a value of 199.2 K/h or almost equal to 200 K/h. Similar dependencies and deviations from the set curve of the heating was obtained for the remaining heating rate. Adopted the setting of the furnace temperature can therefore be considered as correct.

THE RESULTS OF MEASUREMENTS AND CALCULATIONS SCALE ADHE-SION AND STEEL LOSS BY SCALE

The measurements were performed for samples of steel St4S square section 30x30 mm and a height of 50 mm. The heating was conducted in an atmosphere of exhaust gas, the excess air ratio combustion 1,2 to the temperature of the charge surface t = 1250° C. The heating rate M = 100, 200, 300, 400, 500 and 600 K/h were analyzed. First heating measurement was performed only for the

preheating stage, then for a two-stage heating (preheating and soaking) at the same heating rate. Control the composition of the furnace atmosphere (the excess of the combustion air) made by fixing appropriate setting rotameters of air and gas. As part of preliminary tests measurements of the samples mass were carried out and determined their dimensions. The surface area of the samples were calculated. The results of measurements and calculations are presented in Table 1.

The method of the heating carrying out	The preheating rate M, K/h	The initial mass of the sample	The initial height of the sample	The initial size of the side of the sample	The area of the sample A, m ²
		m₀, g	h, mm	a, mm	
The stage of the preheating	100	322,987	50,5	29,0	0,007540
	200	323,276	51,0	29,0	0,007598
	300	322,362	49,5	29,5	0,007582
	400	323,622	51,0	29,0	0,007598
	500	323,892	50,0	30,0	0,007800
	600	321,470	51,0	28,5	0,007439
The process of two-step	100	317,926	50,5	29,0	0,007540
	200	310,792	49,5	29,0	0,007424
	300	321,127	50,5	29,5	0,007700
	400	324,570	51,5	29,5	0,007818
	500	322,622	51,0	29,5	0,007759
	600	320,295	51,0	29,0	0,007598

The next stage of research included measuring the masses of the samples after descaling and total cleaning their surface. The measurement results are compiled in Table 2.

The method of the	The preheating	The sample mass after	The sample mass after
heating carrying out	Rate M, K/h	descaling m ₂ , g	total cleaning m₃, g
	100	322,196	296,963
	200	321,892	299,140
The stage of the preheating	300	320,549	299,865
	400	319,338	308,605
	500	317,718	309,201
	600	315,915	310,768
	100	317,196	275,963
	200	309,493	278,414
The process	300	319,391	294,165
of two-step	400	319,254	299,697
	500	318,211	295,264
	600	317,596	287,632

Table 2 The results of measuring samples mass after descaling and cleaning

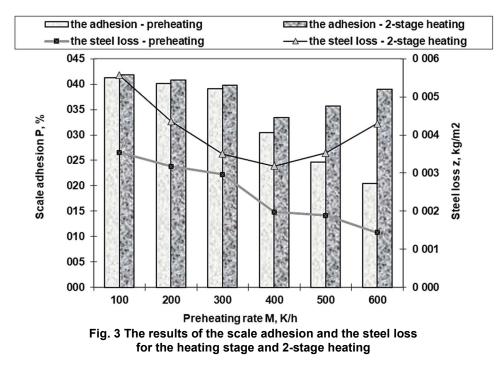
Calculations the steel loss, samples mass after heating and scale adhesion for the analyzed heating rate were made. The results of the calculations are presented in Table 3.

The results of scale adhesion for the heating stage and two-stage heating are shown in Fig. 3.

Graph Fig. 3 shows that with an increase of the heating rate the scale adhesion to a steel substrate for the heating stage is decreasing.

samples mass after heating and scale adhesion					
The method	The preheating	The steel loss by	The sample mass	The scale	
of the heating	rate M, K/h	scale z, kg/m ²	after heating m ₁ , g	adhesion P, %	
carrying out					
The stage of the preheating	100	3,451	358,155	41,24	
	200	3,177	355,892	40,09	
	300	2,967	352,763	39,10	
	400	1,976	343,915	30,40	
	500	1,883	343,745	24,66	
	600	1,439	335,932	20,45	
The process of two-step	100	5,565	374,633	41,79	
	200	4,361	354,546	40,82	
	300	3,502	357,562	39,79	
	400	3,182	358,182	33,44	
	500	3,526	359,592	35,67	
	600	4,299	364,434	39,01	





The lowest value was obtained for M = 600 K/h. After taking into account the period of temperature compensation, the lowest value of adhesion obtained for the heating rate M = 400 K/h. The scale adhesion for individual heating rate reaches a value very similar and the higher, compared with the period of preheating, the higher the heating rate. For speed M = 600 kg/h adhesion has almost doubled, while for M = 100 K/h the increase is almost imperceptible. Simultaneously it can be seen that for two-stage heating adhesion behaves like a loss of steel. Adhesion values in the individual heating stage are determined by the residence time of the charge in temperatures determining the adhesion (above 1000° C). The regularity for selected heating rate is shown in Fig. 4.

Analyzing Fig. 4 it can be concluded that the value of adhesion depends on the residence time at temperatures above 1000°C. Such a thesis also is confirmed by the results of research presented at paper (Halusiak et al., 2012).

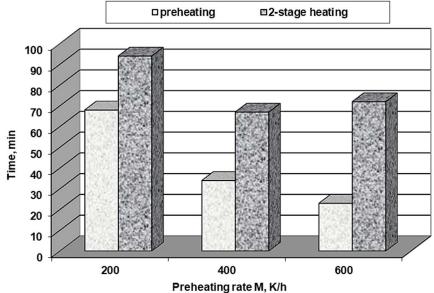


Fig. 4 The residence time of the steel samples at temperatures above 1000°C for the preheating stage and two-stage heating

THE INFLUENCE OF THE HEATING RATE ON ADHESION

Mathematical analysis of the research results were brought to devise depending describing the impact of heating rate on the scale adhesion to a steel substrate for the two-stage heating because preheating stage does not occur in practice without the stage of the temperature leveling. To analyze the effect of heating rate on the adhesion of scale were taken into consideration the following functions:

$$P = 33,345 + 0,14322 \cdot M - 7,07688 \cdot 10^{-4} \cdot M^2 + 1,08486 \cdot 10^{-6} \cdot M^3 - 4,60417 \cdot 10^{-10} \cdot M^4.$$
(14)

$$P = \frac{0.28768}{M^{-1.34}},\tag{15}$$

$$P = 41,269 - 8,09152 \cdot exp \left[-0.5 \cdot \left(\frac{M - 43}{84,61959} \right)^2 \right],$$
(16)

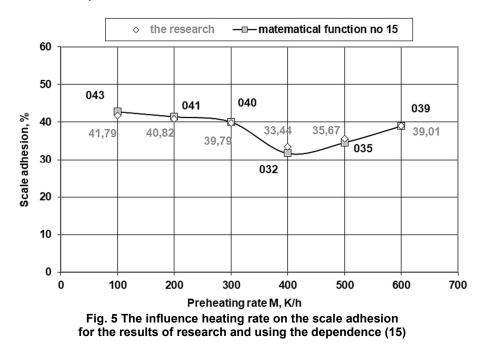
$$P = 42,01556 - 850,83769 \cdot \frac{166,39539}{4 \cdot (M - 438,52238)^2 + (166,39539)^2}$$
(17)

The calculations of the scale adhesion and statistical errors generated using mathematical functions have been made. The values of error statistics are compiled in Table 4.

No dependencies	The correlation coefficient	The sum of squared deviations, Σ²dev.	The average error of the approximation, δ [%]
14	0,935	2,312	6,519
15	0,999	2,116	5,515
16	0,977	1,555	2,336
17	0,960	8,054	89,735

Table 4 The results of calculations of steel loss, samples masses after heating and scale adhesion

Taking into account the value of the statistical error, it was found that the best correlation between the adhesion and the heating rate reproduces relation (15). The dependence of the adhesion on the heating rate using the research results and a mathematical function are shown in Fig. 5.



The calculations indicate statistical error high precision features (14)-(16), but using of the equation (15) is proposed, due to possibility of application issues of scale adhesion to the mathematical considerations.

STATEMENTS AND CONCLUSIONS

The following conclusions can be specify on the basis of theoretical performed analysis and experimental research laboratory:

- a) The heating rate affected on both the steel loss, and the scale adhesion to a steel substrate.
- b) For the heating stage the scale adhesion decreases with increasing of the heating rate.
- c) The application of the two-stage heating, which is necessary to compensate for the temperature difference at the charge section of the steel, results in an increase in adhesion when compared to the preheating stage.
- d) The calculation results may indicate that the rational heating rate will be approximately M = 400K/h.
- e) The scale adhesion to steel substrate retains a similar trend as the steel loss by scale.
- f) The values of the scale adhesion are determined by residence time at temperatures deciding about adhesion (above 1000°C).
- g) The influence of heating rate for 2-stage heating process can be accurately describe mathematical dependencies.
- h) It is proposed to use dependence (15) characterizing by small statistical errors, and yet easy to use in further considerations.

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Abstract.

Before steel materials are subjected to an appropriate heat treatment, they must be preheated. During this process scale is formed, which can be a problem in the proper heat treatment of the steel charge. The preheating process of the steel charge is carried out under specific conditions – parameters. These conditions determine certain properties of scale. One of the most important issues in this context, important from the point of view of the potential removal of scale from the charge surface, is the correlation of the heating process temperature and the adherence of the scale to the steel substrate. The paper presents the results of such research and their discussion. In addition, the methodology for measuring the scale adhesion to steel substrate is presented. Mathematical relations endearing impact of heating rate on the scale adhesion have been developed.

Keywords: heating of charge, preheating rate, heating technology, loss of steel, adhesion of scale