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Table with 2 columns: Item, Value. Contains various technical indicators and their values.

Source: Data provided by the author.

The experimental procedure is similar to that reported in the literature. The samples are prepared by the melt crystallization of the polymer solution. The samples are then subjected to the following procedure:

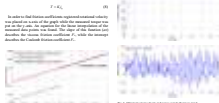


Fig. 1. DSC thermograms of the polymer samples. The glass transition temperature (T_g) and melting temperature (T_m) are indicated by arrows.

6. Laboratory-based experiment
The laboratory-based experiment was conducted using a DSC (DSC Q10, TA Instruments) and a Peltier-type temperature controller (TC-100, TA Instruments). The samples were prepared by the melt crystallization of the polymer solution. The samples were then subjected to the following procedure:

6.1. Sample preparation
The samples were prepared by the melt crystallization of the polymer solution. The samples were then subjected to the following procedure:

6.2. DSC measurement
The DSC measurement was conducted using a DSC (DSC Q10, TA Instruments) and a Peltier-type temperature controller (TC-100, TA Instruments). The samples were then subjected to the following procedure:

6.3. Data analysis
The DSC data were analyzed using the following procedure:

6.4. Results and discussion
The results of the DSC measurement are shown in Figure 1. The glass transition temperature (T_g) and melting temperature (T_m) are indicated by arrows.

6.5. Conclusions
The results of the DSC measurement show that the polymer samples exhibit a glass transition and a melting endotherm.

6.6. Acknowledgments
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