CALCULATION METHOD FOR FLOW BOILING AND FLOW CONDENSATION OF R134A AND R1234YF IN CONVENTIONAL AND SMALL DIAMETER CHANNELS

Dariusz Mikielewicz Blanka Jakubowska Gdansk University of Technology, Poland

ABSTRACT

The paper presents the results of calculations performed using the authors' model to predict the heat transfer coefficient during flow boiling for two refrigerants R134a and R1234yf. The experimental data from various past studies have been collected and the calculations have been conducted for the full range of quality variation and a wide range of mass velocity. The aim of the study was to test the sensitivity of the in-house flow boiling and flow condensation model. The importance of taking into account the surface tension as the parameter exhibiting its importance in case of the flow in minichannels, as well as the influence of the reduced pressure were analysed. The obtained numerical results show good consistency with those recorded in the experiments.

Keywords: heat transfer coefficient, flow boiling, conventional channels, minichannels

NOMENCLATURE

- *B* blowing parameter
- Bo boiling number
- *C* mass concentration of droplets in two-phase core
- *Con* confinement number
- c_p specific heat, J/kgK
- *d* diameter, m
- f, f_r friction factor
- f_{l}, f_{lz} function
- G mass flux, kg/m²s
- h heat transfer coefficient, W/m²K
- *M* molecular weight, kg/kmol
- Nu Nusselt number
- *P* empirical correction
- p pressure, Pa
- Pr Prandtl number
- q heat flux, W/m²
- *r* specific enthalpy of vaporization, J/kg
- *R* two-phase multiplier
- Re Reynolds number
- *s* slip ratio
- T temperature, K
- x quality

GREEK SYMBOLS

- α heat transfer coefficient, W/m²K
 - surface tension
 - thermal conductivity, W/mK
 - density, kg/m³
 - dynamic viscosity, Pa s
- $\xi = \frac{f_r}{4}$ friction factor

SUBSCRIPTS

σ

λ

ρ

- 0 reference case
- *exp* experimental
- *F* Friedel correlation
- g, v vapor
- *kr* critical
- *l* liquid
- *LO* total liquid flow rate
- MS Müller-Steinhagen and Heck correlation
- *Pb* pool boiling
- r reduced
- sat saturation
- *TBP* two-phase boiling
- th theoretical

INTRODUCTION

In the light of increasing interest in refrigerants possessing low global warming potential (GWP), new fluids used in airconditioning and refrigeration units cannot be manufactured with fluorinated greenhouse gases having GWP greater than 150 [1]. Since one of the most popular hydrofluorocarbons (HFC), namely refrigerant R134a, has a 100-year GWP equal to 1430 in low to medium evaporation temperatures [2], it needs to be replaced by more environmentally friendly refrigerants. The need for refrigerant replacements for example in marine air conditioning and refrigeration systems is very urgent, and R134a should be substituted by other environmentally friendly refrigerants in the near future. Previous studies have considered R152a [3] and the natural refrigerant CO₂ [4] as possible replacements for R134a in marine air conditioning systems. It should be borne in mind, however, that R152a is a flammable refrigerant, while for the CO₂ system the working pressure is significantly higher than that for the R134a system, which would lead to significant system modifications and higher costs [5]. Recently, R1234yf has been believed to be a promising candidate as an alternative for R134a [6]. Its ODP is equal to 0, whereas GWP 4 and thermophysical properties are similar to those of R134a. Hence, it meets quite well recent environmental requirements and policies. The main thermophysical properties of R1234yf and R134a are summarized in Table 1.

Park and Jung [8] have reported that R1234yf and R134a have similar nucleate heat transfer coefficients and similar, low toxicity. R1234yf also has mild flammability, compared to R32 and other hydrocarbon refrigerants [7]. Nevertheless, minimizing the refrigerant charge is still important to further reduce the risk related to the flammability of R1234yf, a promising candidate for successful use in systems currently designed for R134a.

There are many available experimental investigations into flow boiling heat transfer of R134a in the literature. Typical experimental ranges of these studies have been collected in Table 2. Unfortunately, the results published for R1234yf are still limited and inconsistent. For instance, Satioh et al. [11] studied the boiling heat transfer of the refrigerant R1234yf flowing in a smooth small-diameter horizontal tube, the inertial diameter of which was 2 mm. From the low to the high vapor quality region the difference between the heat transfer coefficient of R1234yf and R134a is small [9]. Del Col et al. [10] found that there were no significant differences between the flow boiling performance of R1234yf and R134a. On the other hand, Mortada et al. [26] performed an experiment for R1234yf and R134a in a 1.1 mm rectangular channel with mass flux from 20 to 100 kg/m² and heat flux varying from 2 to 15 kW $/m^2$. Their results showed that the heat transfer coefficient for R1234yf is lower by about 40% than that for R134a [9,26]. These results are in opposition to the findings of [10] and [11].

Table 1: Main thermophysical properties of R134a and R1234yf [7].

| | R134a | R1234yf |
|-----------------------------------|-------------|------------------------------------|
| Chemical formula | $C_2H_2F_4$ | CF ₃ CF=CH ₂ |
| Boiling point [°C] | -26 | -29 |
| Critical point [°C] | 102 | 95 |
| Molar mass [kg/kmol] | 102.03 | 114.04 |
| Liquid density at 25 [kg/m3] | 1206.7 | 1091.9 |
| Vapor density at 25 [kg/m3] | 32.350 | 37.925 |
| Liquid enthalpy at 25 [kJ/kg] | 234.55 | 70.076 |
| Vapor enthalpy at 25 [kJ/kg] | 412.33 | 215.44 |
| Liquid therm. cond. at 25 [kW/mK] | 81.134 | 63.585 |
| Vapor therm. cond. at 25 [kW/mK] | 13.825 | 13.966 |
| Liquid Viscosity [µPa s] | 194.89 | 155.45 |
| Vapor Viscosity [µPa s] | 11.693 | 12.291 |
| Surface tension [N/m] | 808×10-5 | 616×10-5 |

In the literature, there are many empirical correlations for modelling of boiling heat transfer, and in the case of such fluids as R134a and R1234yf they have proved rather good consistency with experimental data. Several publications, namely by Ribatski [27], Tibirçá and Ribatski [28], Sardeshpande and Ranade [29], and Alagesan [30] analyse the experimental data for validation of heat transfer coefficient predictions using correlations available in the literature. In the present paper, the experimental results collected from the literature are compared with the predictions of the model [31,32]. Based on the evidence of comparison with the abovementioned experimental data, a correction making use of the reduced pressure effect has been applied to the authors' model to provide feasibly the best consistency of the predictions with the experiments.

THE MODEL

The versatile semi-empirical model to calculate flow boiling and flow condensation due to D. Mikielewicz et al. [31,32] has been tested for a significant number of experimental data and has returned satisfactory results for cases of flow boiling processes in numerous fluids. The fundamental hypothesis of the model is the fact that heat transfer during flow boiling with bubble generation can be modelled as a sum of two contributions constituting the total energy dissipation in the flow, namely the energy dissipation due to shearing flow without bubbles and the dissipation resulting from bubble generation.

Table 2: Range of variation of experimental data for flow boiling of refrigerants R134a and R1234yf, with the corresponding confinement number Con and reduced pressure.

| Authors | Fluid | d [mm] | G [kg/m²s] | q [kW/m²] | T _{sat} [°C] | Con | P_{sal}/P_{kr} |
|----------------------------|---------|-----------|--------------------|----------------|--------------------------|------------------------------|-------------------------------|
| Lu et al. [9] | R1234yf | 3.9 | 200 - 400 | 5.93 - 19.2 | 10 | 0.21931 | 0.12936 |
| | R134a | 3.9 | 200 - 400 | 5.67 - 18.91 | 10 | 0.23401 | 0.10214 |
| Del Col et al. [10] | R1234yf | 1 | 200 - 600 | 30 50 | 31 | 0.73597 | 0.238 |
| Satioh et al. [11] | R1234yf | 2 | 200 300 400 | 6 12 24 | 15 | 0.41411 | 0.15086 |
| | R134a | 2 | 300 | 12 | 15 | 0.44416 | 0.12031 |
| Copetti et al. [12] | R134a | 2.6 | 240 440 | 44 | 22 | 0.3281 | 0.14975 |
| | | | 300 - 435 | 10 20 | 10 | 0.28511 | 0.12936 |
| Kwang-Il Choi et al. [13] | R1234yf | 3 | 300 - 370 | 9 – 25 | 11 | 0.28332 | 0.13347 |
| | | | 300 | 10 | 5 | 0.29391 | 0.11026 |
| | | 1.5 | 500 | 25 | 11 | 0.56664 | 0.13347 |
| Kundu et al. [14] | R134a | 7 | 100 - 400 | 3 – 10 | 7.417 | 0.13213 | 0.09361 |
| Xu et al. [15] | R134a | 2.168 | 725 910 | 19 28.5 | 32.198 27.197 | 0.36848 0.38095 | 0.20201 0.17491 |
| | | 4.065 | 185 295 410 | 18.5 28 | 25.533 20.355 | 0.20534 0.21193 | 0.16653 0.14239 |
| Mancin et al. [16] | R134a | 3.4 | 190 – 755 | 10 25 50 | 30 | 0.23849 | 0.18974 |
| Tibiriçá and Ribatski [17] | R134a | 2.3 | 50 - 600 | 7.5 – 45 | 31 22 41 | 0.35018 0.3709 0.32556 | 0.19525 0.14975 0.25722 |
| Diani et al. [18] | R1234yf | 3.4 | 190 - 940 | 1 – 50 | 30 | 0.21824 | 0.23166 |
| Owhaib et al. [19] | R134a | 1.22 | 200 | 10 34 | 34 | 0.64451 | 0.21251 |
| Shiferaw et al. [20] | R134a | 1.1 | 200 | 10 - 71 | 31 | 0.7322 | 0.19525 |
| Martin-Callizo et al. [21] | R134a | 0.64 | 400 | 5 – 45 | 35 | 1.22388 | 0.21851 |
| Consolini and Thome [22] | R134a | 0.51 | 309 | 12 – 71 | 31 | 1.57926 | 0.19525 |
| Mahmoud et al. [23] | R134a | 0.52 | 280 | 2.6 - 61.7 | 21 | 1.65038 | 0.14524 |
| | | 1.1 | 320 | 12.9 – 102 | 31.327 | 0.73058 | 0.19708 |
| Ong and Thome [24] | R134a | 1.03 | 200 400 1200 | 21.5 - 111.3 | 31 | 0.78196 | 0.19525 |
| | R134a | 1.6 | 300 400 500 | 30 - 80 | 27 32 | 0.51684 0.49997 | 0.1739 0.20088 |
| Anwar et al. [25] | | 0.64 | 330 | 30 | 30 | 1.26697 | 0.18974 |
| | R1234yf | 1.6 | 300 400 | 30 - 80 | 27 32 | 0.47495 0.45617 | 0.21338 0.24447 |

The final version of the model [32] reads:

$$\frac{\alpha_{TBP}}{\alpha_{LO}} = \sqrt{R_{MS}^n + \frac{C}{1+P} \left(\frac{\alpha_{Pb}}{\alpha_{LO}}\right)^2}$$
(1)

In Equation (1) C=1 for flow boiling and C=0 for flow condensation, whereas α_{LO} is the heat transfer coefficient for the liquid only case. It may be determined using, for example, the Dittus – Boelter equation, (for turbulent flow), or Nu=3.66 (for laminar flow). Equation (1) also includes the empirical correction *P* and the modified two-phase multiplier, R_{MS} , due to Müller - Steinhagen and Heck [33]. The modified form of the two-phase multiplier R_{MS} is:

$$R_{MS} = \left[1 + 2 \cdot \left(\frac{1}{f_1} - 1\right) \cdot x \cdot Con^m\right] \cdot (1 - x)^{\frac{1}{3}} + x^3 \cdot \frac{1}{f_{1z}}$$
(2)

It should be noted that the two-phase multiplier R_{MS} in Equation (2) is raised to the power n, where n = 0.76 for turbulent flows, and n = 2 for laminar flows. Function f_1 in Equation (3) is defined as the ratio of the pressure drop in the single-phase flow of liquid to the pressure drop in the single-phase flow of gas, whereas f_{1z} denotes the ratio of the single-phase flow heat transfer coefficient in vapour to the single-phase heat transfer coefficient for liquid, respectively. For turbulent flows these functions can be determined from relations: $f_1 = (r_L/r_G)(m_L/m_G)^{0.25}$, $f_{1z} = (m_G/m_L)(l_L/l_G)^{1.5}(c_{pL}/c_{pG})$, while for laminar flows: $f_1 = (r_L/r_G)(m_L/m_G)^{-1}$ and $f_{1z} = (l_G/l_L)$. Furthermore, the exponent m=0 for flow in conventional channels, and m=-1 for flow in minichannels. The empirical correction P in Equation (1) should be calculated as:

$$P = 2.53 \cdot 10^{-3} \cdot \text{Re}^{1.17} \cdot Bo^{0.6} \cdot (R_{MS} - 1)^{-0.65}$$
(3)

The pool-boiling heat transfer coefficient α_{pb} (1) can be calculated using a generalized model due to Cooper [34]. This model describes the heat transfer coefficient in the fluid in terms of the reduced pressure, molecular weight, and applied wall heat flux. The Cooper equation which describes the pool-boiling heat transfer coefficient has the form:

$$\alpha_{Pb} = A \cdot p_r^{0.12} \cdot (-\log p_r)^{-0.55} \cdot M^{-0.5} \cdot q^{\frac{2}{3}}$$
(4)

The accuracy of model predictions was expected to be improved by some modifications introduced to the empirical correction *P*. The modified empirical correction *P* yields:

$$P = 2.53 \cdot 10^{-3} \cdot \text{Re}^{1.17} \cdot Bo^{0.6} \cdot (R_{MS}^* - 1)^{-0.65} \cdot \left(\frac{p_{sat}}{p_{kr}}\right)^a$$
(5)

 R_{MS}^{*} in Equation (5) is the two-phase multiplier, calculated using the original version of Muller-Steinhagen and Heck correlation [33]. In using Equation (2), the exponent is always m=0. Exponent *a* was adjusted to the available data

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bank for R134a and R1234yf presented in Table 2. Another modification of Equation (1) was the change to the value of the exponent *m* in Equation (2). The sensitivity of the developed model of two-phase flow multiplier and the nonadiabatic effect were tested in calculations. For that purpose, two models were introduced into Equation (1), namely the Muller – Steinhagen and Heck correlation [33]. Additionally, the calculation procedure includes the so called blowing parameter *B*, which is responsible for evaluation of the nonadiabatic effects appearing due to shear stress modification on the liquid-vapour interface [32].

The modified two-phase multiplier R_B , inclusive of nonadiabatic effects, has the following form [32]:

$$R_{B} = \begin{cases} R_{MS} \cdot \left(1 - \frac{B}{2}\right) & \text{for } 0.1 < x \le 1 \\ R_{MS} \sqrt{1 + \left(\frac{8 \cdot \alpha_{Pb} \cdot d_{w}}{\lambda_{l} \cdot \text{Re} \cdot \text{Pr} \cdot \xi_{0} \cdot R_{MS}}\right)^{2}} & \text{for } 0 \le x \le 0.1 \end{cases}$$
(6)

In Equation (6) the two-phase multiplier should be calculated by using the modified Muller-Steinhagen and Heck correlation, Equation (2). The blowing parameter in Equation (6) is defined as [32]:

$$B = \frac{2 \cdot q \cdot \frac{\rho_l}{\rho_v}}{f_r \cdot G(s-1) \cdot r}$$
(7)

In Equation (7), *s* is the slip ratio which can be determined from the Zivi relation [32].

$$s = \sqrt[3]{\frac{\rho_l}{\rho_v}} \tag{8}$$

As a result of application of correction (6), a modified heat transfer model is obtained. This model has been adopted for calculations in the present work:

$$\frac{\alpha_{TBP}}{\alpha_{LO}} = \sqrt{R_B^n + \frac{C}{1 + 2.53 \cdot 10^{-3} \cdot \text{Re}^{1.17} \cdot Bo^{0.6} \cdot (R_{MS}^* - 1)^{-0.65} \cdot p_r^a} \left(\frac{\alpha_{Pb}}{\alpha_{LO}}\right)^2}$$
(9)

In the following part, the basic model and its subsequent modifications, which have been selected for discussion, is analysed with respect to predictions of heat transfer coefficient. The exponent *a* in the modified two-phase flow multiplier in Equation (5) was adjusted to the available data bank for flow boiling of R134a and R1234yf.

THE RESULTS

To distinguish between the conventional channels and minichannels the criterion due to Kew and Cornwell [35] has been used, which is based on the so-called confinement number *Con*, defined as:

$$Con = \frac{1}{d} \sqrt{\frac{\sigma}{d(\rho_l - \rho_v)}}$$
(10)

It has been postulated that when the confinement number *Con* is greater than 0.5 then the flow exhibits the properties of the flow in minichannel, in which the surface tension plays a dominant role. Presented below is a selection of studies which have been considered in the present study.

Based on the Kew and Cornwell [35] criterion, the available data bank was divided into conventional size channels and minichannels. Amongst the collected data, the criterion of minichannels, i.e. *Con* > 0.5, is fulfilled only by the research due to De Col et al. [10] and Kwang-Il Choi [13] for R1234yf, while for R134a this criterion is fulfilled by research due to Owhaib et al. [19], Shiferaw et al. [20], Martin-Callizo et al. [21], Consolini and Thome [22], Mahmoud et al. [23], and Anwar et al. [25]. It can therefore be concluded that for both R134a and R1234yf, the transition from conventional size channels to minichannels takes place at a channel diameter smaller than 1.5mm. The relations of the confinement number Con to the saturation temperature for both refrigerants are presented in Figures 1 and 2. The values of the confinement number Con for the analysed refrigerant, together with the reduced pressure values and the range of variation of experimental parameters, are collated in Table 2. The analysis of the parameters in Table 2 indicates the fact that the experimental research data collected for scrutiny cover a full range of quality variation and a relatively wide range of mass velocity.



Figure 1: Confinement number Con vs. saturation temperature for R134a



Figure 2: Confinement number Con vs. saturation temperature for R1234yf

Figures 3 to 6 show the results of heat transfer coefficient calculations for R134a and R1234yf by using the earlier mentioned method based on Equation (9). In the adopted formulation, the surface tension effects are included into the analysis by considering the effect of the confinement number in equations through Con. The model version applicable to minichannels was used if Con>0.5, and that applicable to convectional size channels when Con<0.5. The modification of the empirical correction described by Equation (3) includes the effect of reduced pressure (p_{sat}/p_{kr}) . The new version of the correction *P* is applied in Equation (5), where the reduced pressure is raised to the power *a*. The value of the exponent a was adjusted using the regression analysis of functions of several variables. The accomplished calculations indicate that best consistency is obtained if a = 4 for each case. The information about mean average deviation and correlation factors is given in Table 3. The values of correlation factors are not very high, which indicates relatively high dispersity of the experimental data.



Figure 3: Comparison of test results, α_{gep} with predictions obtained using Equation (9), α_{a} and a = 4 for R134a



Figure 4: Comparison of the ratio of experimental values of α_{csp} to those obtained using Equation (9), α_{th} as function of quality, and a = 4 for R134a



Figure 5: Comparison of test results, a_{exp} with predictions obtained using Equation (9), a_{th} and a = 4 for R1234yf



Figure 6: Comparison of the ratio of experimental values of α_{exp} to those obtained using Equation (9), α_{th} as function of quality, and a = 4 for R1234yf

Table 3: Values of exponent a and MAD

| Fluid | Model | R^2 | MAD [%] |
|---------|----------|--------|------------|
| R134a | Model II | 0.1536 | 25.35 |
| R1234yf | Model II | 0.4522 | 16.06 |

Based on the presented results of calculations, which were obtained using the versions of the heat transfer model described by Equation (9), with and without the account of reduced pressure $(p_{sat}/p_{kr})^a$ from Equation (5), it can be concluded that taking into account the effect of reduced pressure improves the convergence with the experimental data, consequently giving smaller values of the mean absolute deviation and higher of the correlation coefficient.

CONCLUSIONS

The paper presents the analysis of the results of flow boiling and flow condensation calculations using a model developed earlier to study experimental data for flow boiling of R134a and R1234yf. The model was studied in several ways: it was used either in the original formulation or in a modified version, which included the reduced pressure effect in the empirical correction *P* (value of exponent *a* was modelled). The results show that the effect of reduced pressure does not significantly change the performance of the original model. The results of calculations show the best compliance with the experimental data when applying the blowing parameter in correlation for each case, i.e. for R134a and R1234yf, and also with and without the account of reduced pressure (p_{sal}/p_{k})^{*a*} from Equation (5).

In authors' opinion, the proposed method to calculate the flow boiling heat transfer coefficient is a reliable tool in designing heat exchangers. Among other possibilities, it provides opportunities for analysing other materials which would be more suitable for marine applications, as well as new manufacturing techniques which can be fully automated [36,37]. At present, heat exchangers are normally equipped with stainless steel tubes. In future applications, some effort should be made to use plastics [38-40], which do not exhibit susceptibility to corrosion and are cheaper.

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CONTACT WITH THE AUTHOR

Dariusz Mikielewicz, *e-mail: Dariusz.Mikielewicz@pg.gda.pl,*

Gdansk University of Technology, Faculty of Mechanical Engineering, Department of Energy and Industrial Apparatus, ul. Narutowicza 11/12, 80-233 Gdansk **POLAND**

> Blanka Jakubowska e-mail: blanka.jakubowska@pg.gda.pl

Gdansk University of Technology, Faculty of Mechanical Engineering, Department of Energy and Industrial Apparatus, ul. Narutowicza 11/12, 80-233 Gdansk **POLAND**