

## MECHANICAL PROPERTIES AND CORROSION BEHAVIORS OF AGED Ti-4Mo-4Cr-X (X = Sn, V, Zr) ALLOYS FOR METALLIC BIOMATERIALS

The purpose of this study was to investigate the mechanical properties of beta type aged Ti-4Mo-4Cr-X (X = V, Sn, Zr) quaternary alloy for use as a cardiovascular stent. Titanium (Ti) alloys were fabricated using a vacuum arc remelting furnace process. To homogenize the specimens of each composition and remove the micro segregation, all cast specimens were subjected to homogenization at 850°C for 4 h, which was 100°C higher than the  $\beta$ -transus temperature of 750°C. The tensile strength and elongation of the aged Ti-4Mo-4Cr-X (X = V, Sn, Zr) alloys were increased as compared to the homogenized alloys. In addition, many  $\alpha/\beta$  interface boundaries formed after aging treatment at 450°C, which acted as inhibitors of strain and caused an increase in tensile strength. The elongation of Ti-4Mo-4Cr-X alloys consisting of  $\alpha + \beta$  phases after aging treatment was improved by greater than 30%. Results of a potentiodynamic polarization test showed that the lowest current density of Ti-4Mo-4Cr-4Sn with  $1.05 \times 10^{-8} \text{ A/cm}^2$  was obtained. The present Ti-4Mo-4Cr-X alloys showed better corrosion characteristics as compared to the 316L stainless steel and L605 (Co-Cr alloy) cardiovascular stent alloys.

*Keywords:* Beta Ti-alloy, Metallic biomaterials, Aging, Potentiodynamic polarization

### 1. Introduction

Research on the development of titanium (Ti)-based stent materials for blood vessels has been rarely conducted. Ti-based alloys for bare-metal stents are extremely attractive because of their high strength, low elastic modulus, excellent corrosion resistance, and superior biocompatibility [1]. An ideal bare metal stent has a low profile, good expandability ratio, sufficient radial hoop strength, negligible recoil, and sufficient flexibility [2].

Ti alloys are designed by alloying quantities of  $\alpha$ - and  $\beta$ -stabilizing elements. Thus, the fractions of  $\alpha$  and  $\beta$  phases are determined by the alloying elements [3]. Beta-phase stabilizers including Mo, Cr, Nb, Mn, Co, and Fe have been studied for biomedical applications. The molybdenum equivalence ( $\text{Mo}_{\text{eq}}$ ) in the range of 12-15 wt.% has been reported as an optimal combination of strength and toughness [4].

Ti alloy performance is strongly dependent on the controlled thermomechanical treatment [5]. Further plastic deformation, including texture evolution, and aging treatment result in grain refinement and second-phase precipitation, which are required for the strength-ductility trade-off [6]. This study investigated the mechanical properties and corrosion behaviors in Ringer's

solution of a beta-type aged Ti-4Mo-4Cr-X (X = V, Sn, Zr) quaternary alloy for use as a cardiovascular stent.

### 2. Experimental

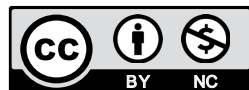
Ingots of Ti-4Mo-4Cr-X (X = 2, 3, 4 wt.% V, 2, 3, 4 wt.% Sn, and 2, 3, 4 wt.% Zr) quaternary alloys were fabricated through a vacuum arc remelting (VAR, ACE VACUUM, AVA-1500, Korea). Commercially pure Ti chips (ASTM CP Grade II), Mo bars (99.8 wt.%), Cr chips (99.9 wt.%), V sheets (99.9 wt.%), Sn balls (99.9 wt.%), and Zr sheets (99.9 wt.%) were arc melted in a water-cooled copper hearth with a tungsten electrode. The ingots were remelted four times under an argon atmosphere and hot forged with a thickness reduction of 35% to ensure chemical homogeneity. All cast ingots were then homogenized at 850°C for 4 h, which was 100°C higher than the  $\beta$ -transus temperature of 750°C. In addition, aging treatment was conducted at 450°C for 16 h.

The phase constitutions of the alloys were examined by X-ray diffraction (XRD, PANalytical, X'Pert pro, Netherland) analysis using Cu-K $\alpha$  radiation over  $2\theta$  range from 30-90° at

<sup>1</sup> CHONNAM NATIONAL UNIVERSITY, SCHOOL OF MATERIALS SCIENCE & ENGINEERING, GWANGJU, REPUBLIC OF KOREA

<sup>2</sup> 21CENTURY MEDICAL CO. LTD., GWANGJU, KOREA

\* Corresponding author: kmlee@jnu.ac.kr



an accelerating voltage of 40 kV, a current of 250 mA, and a scanning speed of  $2^\circ/\text{min}$ . Tensile specimens were manufactured in accordance with the ASTM E8 standard. A tensile test was conducted using a universal material tester (Shimadzu: AG-100KNIC) at a load of 20,000 N under a tensile load of 10 mm/min. The fractography after the tensile test was examined using a scanning electron microscope (S-4700, HITACHI, Japan). Electrochemical experiments were performed on a flat cell corrosion tester (PARSTAT 2273, USA) at a temperature of  $37 \pm 1^\circ\text{C}$ . A three-electrode cell was used for potentiodynamic polarization tests, where the reference electrode was a silver–silver chloride electrode, the counter electrode was made of a platinum plate, and the specimen was the working electrode. All experiments were conducted at a constant scan rate of 0.25 mV/s, initiated at  $-250$  mV below the open-circuit potential. The working electrolyte was Ringer's physical solution.

### 3. Results and discussion

Fig. 1 shows the XRD patterns of the aged (a) Ti-4Mo-4Cr- $x$ V ( $x = 2, 3, 4$ ) alloys, (b) Ti-4Mo-4Cr- $x$ Sn ( $x = 2, 3, 4$ ) alloys, and (c) Ti-4Mo-4Cr- $x$ Zr ( $x = 2, 3, 4$ ) alloys. The Ti-4Mo-4Cr-(2, 3, 4) V alloys showed that  $\beta$ -phase peaks formed in the (110), (200) and (211) planes. It was also observed that  $\alpha$ -phase peaks formed in the (100), (002), (102) and (110) planes in the alloys containing 2 wt.% and 3 wt.% V. The Ti-4Mo-4Cr-4V alloy nearly showed stable  $\beta$ -phase peaks because of a higher value of  $\text{Mo}_{\text{eq}}$  (12.2). The beta phase stability in Ti-4Mo-4Cr- $x$ V ( $x = 2, 3, 4$ ) alloys also increased with increasing V content. By contrast, the  $\alpha$ -phase peaks of aged Ti-4Mo-4Cr- $x$ Zr ( $x = 2, 3, 4$ ) alloys were found to be higher than those of homogenized specimens, suggesting that precipitation of  $\alpha$  phases increased with aging treatment. The precipitation of  $\alpha$  phases could be attributed to the existence of  $\alpha''$  phases at the grain boundaries, which acted as precursor nucleation sites for the stable  $\alpha$  phases.

Fig. 2 shows the stress-strain curves after tensile testing of Ti-4Mo-4Cr- $X$  ( $X = \text{V, Sn, Zr}$ ), which were homogenized at  $850^\circ\text{C}$  for 4 h and subsequently aging treated at  $450^\circ\text{C}$  for 16 h. As shown in Fig. 2, the tensile strength of the aged Ti-4Mo-4Cr- $X$  ( $X = \text{V, Sn}$ ) alloys increased with increasing V or Sn content. The maximum tensile strength for the Ti-4Mo-4Cr-4Sn alloy was approximately 1554 MPa with an elongation of 38%. The tensile strength of Ti-4Mo-4Cr- $x$ Zr ( $x = 2, 3, 4$ ) alloys decreased with an increasing amount of Zr. The high tensile strength and elongation for all aged specimens can attribute to many  $\alpha$  phases formed after the aging treatment, which act as inhibitors of dislocation motion in interface boundaries between the  $\beta$  and  $\alpha$  phases. The co-existence of  $\alpha$  phases could also induce a slightly improved elongation in the aged Ti-4Mo-4Cr- $X$  ( $X = \text{V, Sn, Zr}$ ) alloys. The highest value of elongation multiplied tensile strength ( $\text{MPa} \times \%$ ) was  $5.9 \times 10^4$  for the Ti-4Mo-4Cr-4Sn alloy, which was increased up to 22% after aging processing. As known Ti-Sn equilibrium phase diagram, the Sn containing alloy system has some intermetallic phases such as  $\text{Ti}_3\text{Sn}$  or  $\text{Ti}_2\text{Sn}$ , which cause

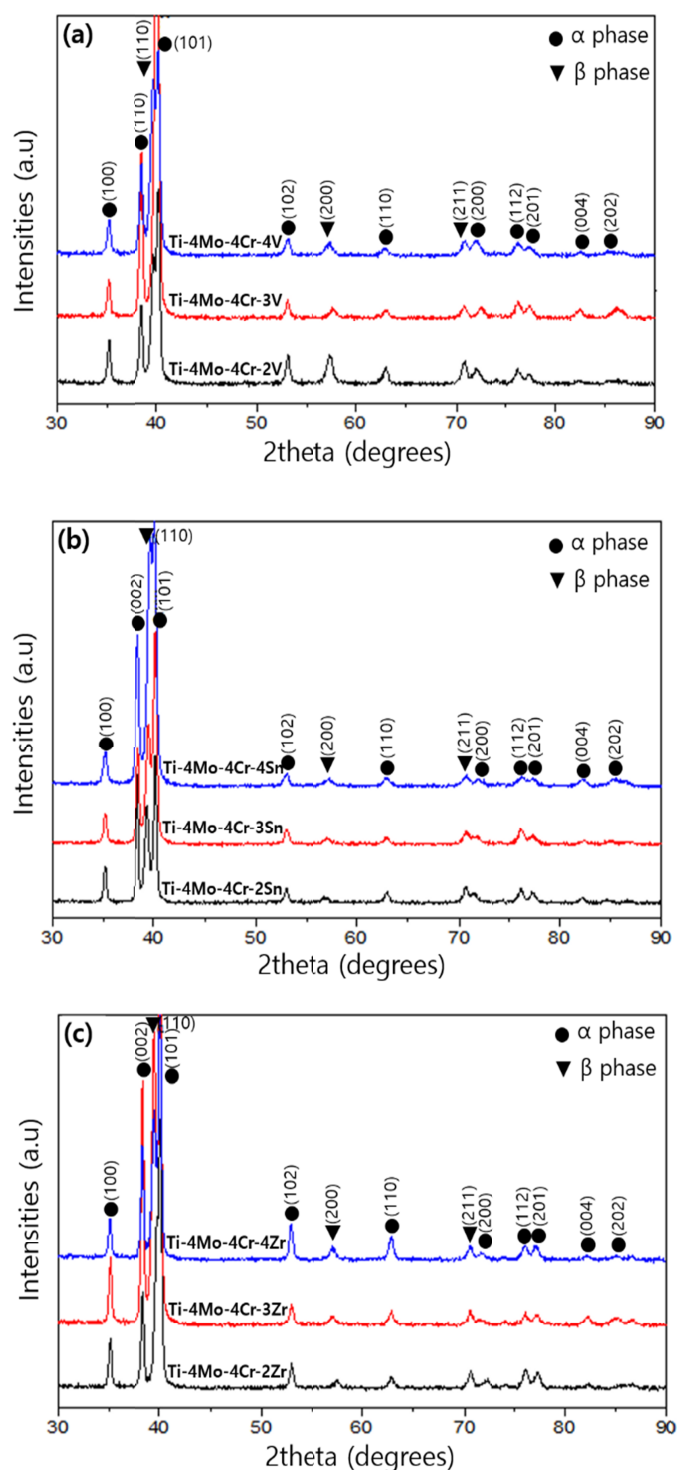


Fig. 1. XRD patterns of (a) Ti-4Mo-4Cr- $x$ V ( $x = 2, 3, 4$ ) alloys, (b) Ti-4Mo-4Cr- $x$ Sn ( $x = 2, 3, 4$ ) alloys, and (c) Ti-4Mo-4Cr- $x$ Zr ( $x = 2, 3, 4$ ) alloys, which were homogenized at  $850^\circ\text{C}$  for 4 h and subsequent aging treated at  $450^\circ\text{C}$  for 16 h

high tensile strength. Consequently, the aged Ti-4Mo-4Cr-4Sn alloy could be considered as an alloy candidate for the strength-ductility trade-off.

Fig. 3 shows SEM fractography of the aged (a) Ti-4Mo-4Cr-3V, (b) Ti-4Mo-4Cr-3Sn and (c) Ti-4Mo-4Cr-3Zr alloys after the tensile test. All aged Ti-4Mo-4Cr-3 $X$  ( $X = \text{V, Sn, Zr}$ ) alloys after the tensile test showed a ductile fracture with

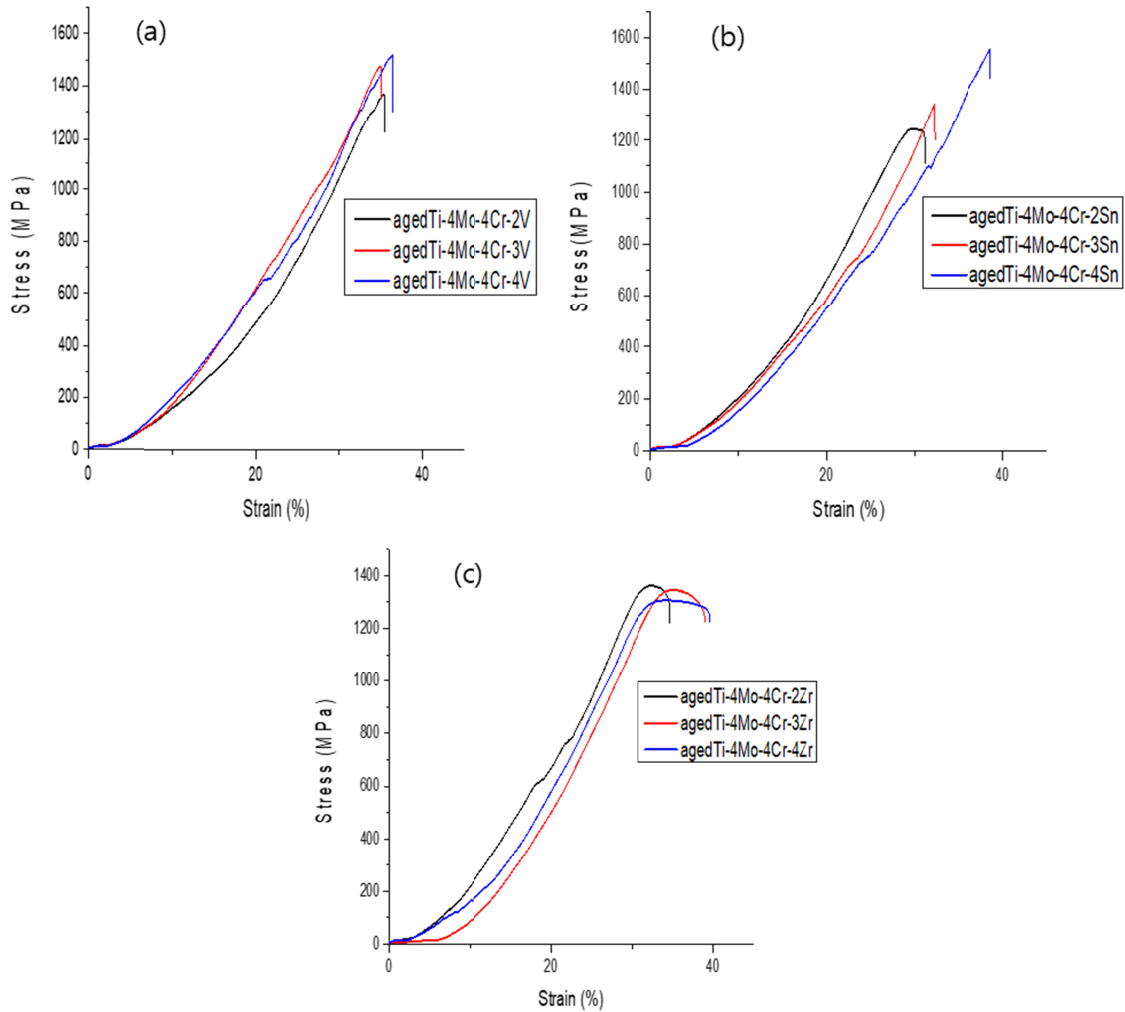


Fig. 2. Stress-strain curves after tensile tests of (a) Ti-4Mo-4Cr- $x$ V ( $x = 2, 3, 4$ ) alloys, (b) Ti-4Mo-4Cr- $x$ Sn ( $x = 2, 3, 4$ ) alloys, and (c) Ti-4Mo-4Cr- $x$ Zr ( $x = 2, 3, 4$ ) alloys, which were homogenized at 850°C for 4 h and subsequent aging treated at 450°C for 16 h

an elongation greater than 30% (31-39%). In Fig. 3, many fine dimple structures of less than 10  $\mu$ m were observed in the alloy.

Fig. 4 shows the potentiodynamic polarization curves of the Ti-4Mo-4Cr- $X$  ( $X = V, Sn, Zr$ ) alloys as a function of the addition of an alloying element. Corrosion behaviors relating to biocompatibility are the main factors in a cardiovascular stent. In general,  $E_{corr}$  and  $i_{corr}$  represent the corrosion potential and corrosion current density, respectively. The lowest current density ( $1.05 \times 10^{-8}$  A/cm<sup>2</sup>) was observed in the Ti-4Mo-4Cr-4Sn alloy, whereas the specimen with the Ti-4Mo-4Cr-2Zr alloy showed the highest corrosion current density of  $2.33 \times 10^{-7}$  A/cm<sup>2</sup>. As shown in the polarization graph, a passive layer on all the specimens was formed at a slow rate. The values of  $E_{corr}$  and  $i_{corr}$  for Ti-4Mo-4Cr- $X$  ( $X = V$  or Sn) were found to be in the range of  $-0.032$  and  $-0.317$  V and the order of  $10^{-8}$  A/cm<sup>2</sup>, respectively. Basically, the addition of Mo and Cr elements to Ti alloys results in an improved corrosion resistance because of the formation of a passive films of TiO<sub>2</sub> and MoO<sub>3</sub> [7] and the formation of a chromium oxide-rich surface film in a fluoride-containing saline solution [8]. The Ti-4Mo-4Cr- $X$  ( $X = V$  or Sn) alloys also showed an improved corrosion resistance with the exception of

the Ti-4Mo-4Cr- $x$ Zr ( $x = 2, 3, 4$ ) alloys. However, in comparison with those of the available cardiovascular stent materials such as 316L stainless steel and L605 alloy, the corrosion resistance of the beta-type aged Ti-4Mo-4Cr- $X$  ( $X = V, Sn, Zr$ ) quaternary alloys appeared to be superior.

#### 4. Conclusion

In this study, the mechanical properties and corrosion behaviors in Ringer's solution of a beta-type aged Ti-4Mo-4Cr- $X$  ( $X = V, Sn, Zr$ ) quaternary alloy were investigated. The following results were derived from this study.

1. 4<sup>th</sup> alloying elements, V as a  $\beta$ -phase stabilizing element and Sn or Zr as a  $\alpha$ -phase stabilizing element, significantly affected the degree to which the volume of each phase could exist in quaternary Ti-4Mo-4Cr- $X$  ( $X = V, Sn, Zr$ ) alloys. The formation of the  $\alpha$  phase after aging treatment contributed to an increase in the tensile strength and elongation of the alloy because the interface boundaries between the  $\beta$  and  $\alpha$  phases can act as inhibitors of dislocation motion.

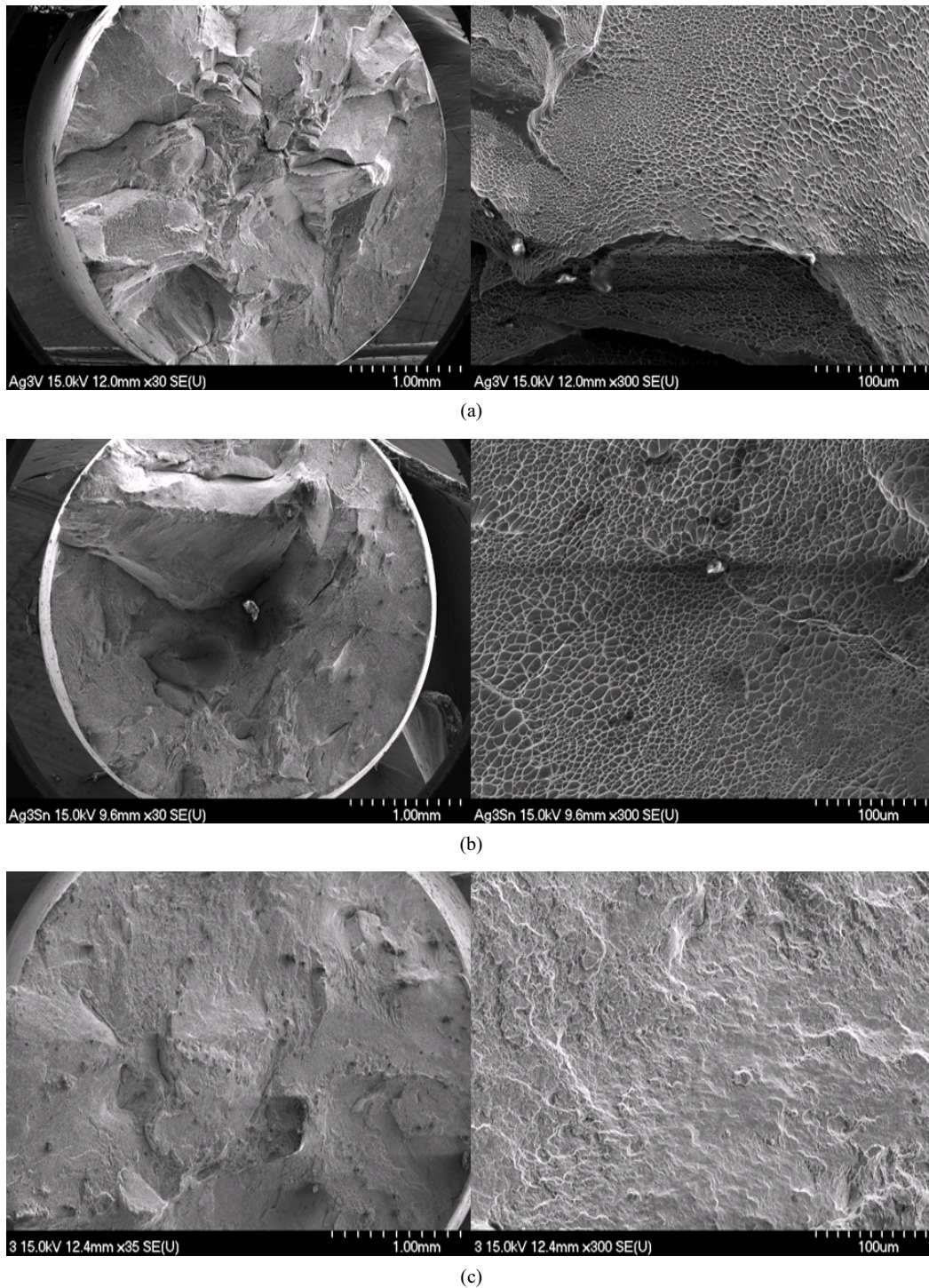


Fig. 3. SEM fractography of the aging treated Ti-4Mo-4Cr-X alloys after tensile tests: (a) Ti-4Mo-4Cr-3V, (b) Ti-4Mo-4Cr-3Sn and (c) Ti-4Mo-4Cr-3Zr

2. The tensile strength and elongation of the aged Ti-4Mo-4Cr-X (X = V, Sn, Zr) alloys increased as compared to the homogenized Ti-4Mo-4Cr-X (X = V, Sn, Zr) alloys. The aged Ti-4Mo-4Cr-4Sn alloy can be considered a Ti-alloy candidate for the strength-ductility trade-off.
3. Result of a polarization potential test revealed that the lowest current density of Ti-4Mo-4Cr-4Sn with  $1.05 \times 10^{-8}$  A/cm<sup>2</sup> was obtained. The studied Ti-4Mo-4Cr-X (X = V, Sn, Zr) alloys showed better corrosion characteristics as

compared to the available cardiovascular stent materials such as 316L stainless steel and L605 (Co-Cr) alloy.

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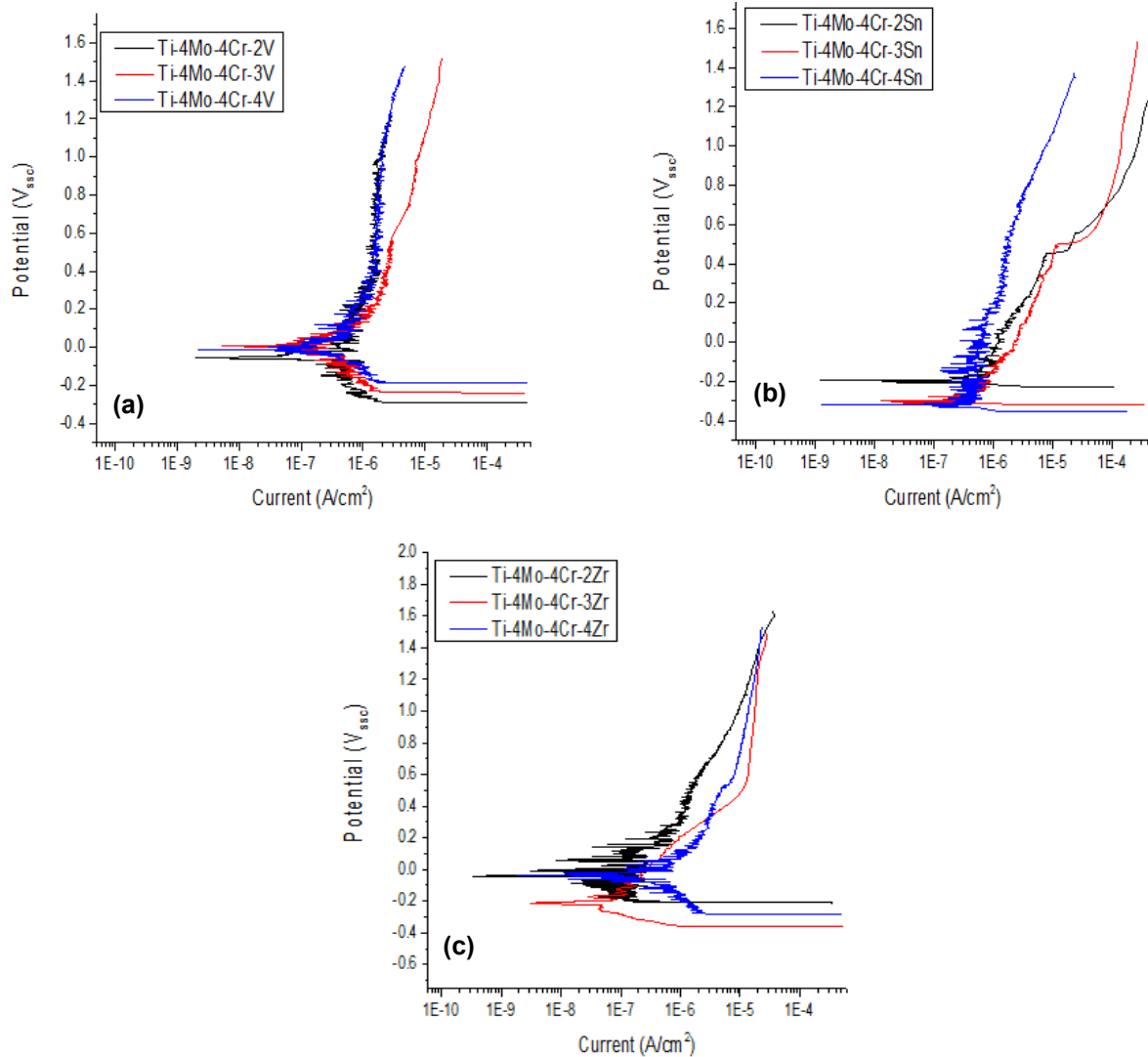


Fig. 4. Potentiodynamic polarization curves of (a) Ti-4Mo-4Cr-xV ( $x = 2, 3, 4$ ) alloys, (b) Ti-4Mo-4Cr-xSn ( $x = 2, 3, 4$ ) alloys, and (c) Ti-4Mo-4Cr-xZr ( $x = 2, 3, 4$ ) alloys

TABLE 1

$E_{\text{corr}}$  and  $i_{\text{corr}}$  values obtained at potentiodynamic polarization curves of Fig. 4

	Ti-4Mo-4Cr-xV (wt.%)			Ti-4Mo-4Cr-xSn (wt.%)			Ti-4Mo-4Cr-xZr (wt.%)		
	2	3	4	2	3	4	2	3	4
$E_{\text{corr}}(\text{V}_{\text{SSC}})$	-0.053	-0.039	-0.032	-0.196	-0.264	-0.317	-0.002	-0.170	-0.044
$I_{\text{corr}}(\text{A}/\text{cm}^2)$	$2.79 \times 10^{-8}$	$5.33 \times 10^{-8}$	$8.75 \times 10^{-8}$	$4.64 \times 10^{-8}$	$1.62 \times 10^{-8}$	$1.05 \times 10^{-8}$	$2.33 \times 10^{-7}$	$2.74 \times 10^{-8}$	$1.04 \times 10^{-7}$
Control	316L Stainless Steel					L605 Alloy (Co-20Cr-15W-10Ni) (wt.%)			
$E_{\text{corr}}(\text{V}_{\text{SSC}})$	-0.564					-0.751			
$I_{\text{corr}}(\text{A}/\text{cm}^2)$	$1.30 \times 10^{-7}$					$2.29 \times 10^{-5}$			

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